

Answer all the questions.

5. Methanoic acid, HCOOH , is a carboxylic acid that is present in some stings.

(a) What is the formula of the functional group that is responsible for the characteristic properties of carboxylic acids?

[2]

(b) Methanoic acid is usually referred to simply as formic acid.

Limestone is made of calcium carbonate, which is insoluble in water.

Carboxylic acids react with carbonates in a similar way to other acids such as hydrochloric acid.



(c) Complete and balance this symbol equation for the reaction between calcium carbonate and methanoic acid.



[2]

(d) Calcium carbonate is insoluble so it does not react with water.

When calcium carbonate is heated, it reacts with methanoic acid, calcium methanoate forms.

The reaction with methanoic acid removes the calcium carbonate in limestone. Suggest a property of calcium methanoate that can explain why this happens.

[1]

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