

GRAPHING TRENDS OF THE PERIODIC TABLE

CHEMISTRY

NAME: _____

DATE: _____

PERIOD: _____

Graphing Periodic Table Trends

Directions: Using the provided graph paper, graph each of the elemental properties using a line graph. Place the property on the Y-axis and the Atomic number on the X-axis. Draw a dark, solid, vertical line on each graph where each row on the periodic table ends. (i.e. after atomic number 10). Use the graphs to analyze the trends in these properties and to answer the questions on the back.

| Atomic Radius (pm) | | | | | | | |
|--------------------|-----|-----------------|-----|----------------|-----|----------------|-----|
| Lithium - 3 | 157 | Sodium - 11 | 190 | Potassium - 19 | 243 | Rubidium - 27 | 265 |
| Beryllium - 4 | 112 | Magnesium - 12 | 145 | Calcium - 20 | 194 | Strontium - 28 | 219 |
| Boron - 5 | 87 | Aluminum - 13 | 118 | Gallium - 31 | 136 | Indium - 49 | 156 |
| Carbon - 6 | 67 | Silicon - 14 | 111 | Germanium - 52 | 125 | Tin - 50 | 140 |
| Nitrogen - 7 | 56 | Phosphorus - 15 | 98 | Arsenic - 33 | 114 | Antimony - 51 | 133 |
| Oxygen - 8 | 48 | Sulfur - 16 | 88 | Selenium - 34 | 103 | Tellurium - 52 | 123 |
| Fluorine - 9 | 42 | Chlorine - 17 | 79 | Bromine - 35 | 94 | Iodine - 53 | 115 |

| Ionization Energy (kJ/mol) | | | | | | | |
|----------------------------|-----|----------------|-----|----------------|-----|----------------|-----|
| Lithium - 3 | 520 | Sodium - 11 | 496 | Potassium - 19 | 419 | Rubidium - 27 | 403 |
| Beryllium - 4 | 900 | Magnesium - 12 | 738 | Calcium - 20 | 590 | Strontium - 28 | 550 |

Atomic Radius



Periodic Trends Graphing And Questions Key

Christian Drosten

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Science Units for Grades 9-12 Randy L. Bell, Joe Garofalo, 2005 Sample topics include cell division virtual dissection earthquake modeling the Doppler Effect and more

Chemistry in the Community (Enhanced Core Four) American Chemical Society, 2006-02-15

ChemDiscovery Student Guide Olga I. Agapova, 2002

Chemistry in the Community. American Chemical Society, 2002 This volume has relevance to a wide number of courses giving a hands on introduction to chemistry in relation to community issues rather than around specific chemical concepts

Hands-On General Science Activities With Real-Life Applications Pam Walker, Elaine Wood, 2008-04-21 In this second edition of Hands On General Science Activities with Real Life Applications Pam Walker and Elaine Wood have completely revised and updated their must have resource for science teachers of grades 5-12 The book offers a dynamic collection of classroom ready lessons projects and lab activities that encourage students to integrate basic science concepts and skills into everyday life

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Advanced Chemistry Michael Clugston, Rosalind Flemming, 2000-06-08 Carefully researched by the authors to bring the subject of chemistry up to date this text provides complete coverage of the new A and AS level core specifications The inclusion of objectives and questions make it suitable for self study

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Holt Chemistry Ralph Thomas Myers, 2004

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