| Name   | Date           | Class            |
|--|----------------|------------------|
| Control of the contro | L. Partie B.C. | Page 18-pt 18-pt |



#### THE PERIODIC TABLE

#### Chapter 6 Review Package - Part 1

#### A. Matching

Match each description in Column B with the correct term in Column A. Write the letter of the correct description on the line.

|     | Column A               |     | Column B   |
|-----|------------------------|-----|--|
| 37  | metal                  | 46. | positively charged ion   |
| 2.  | halogen                | b.  | an element whose highest occupied $s$ or $\rho$ sublevels are partially filled             |
| 3   | inner transition metal | ¢.  | the energy needed to remove an electron from an atom in the gaseous state                  |
| 4.  | representative element | d.  | good conductor of heat and electric current  |
| 5   | ionization energy      | e.  | an element whose highest occupied $s$ sublevel and a nearby $d$ sublevel contain electrons |
| 6.  | alkaline earth metal   | ď.  | any nonmetal in Group 7A   |
| 7.  | cation                 | g.  | the tendency of an atom to attract electrons when the atom is in a compound                |
| 8.  | electronegativity      | h.  | an element in which the highest occupied $s$ and $\rho$ sublevels are filled               |
| 9.  | transition metal       | 1.0 | an element in Group 2A   |
| 10. | noble gas              | J.  | an element whose highest occupied s sublevel and a nearby f sublevel contain electrons     |

- 11. In the periodic table, there is a periodic pattern in the physical and chemical properties of elements when they are arranged in order of a. increasing atomic mass.
  - b. increasing electronegativity.
  - e. increasing atomic radius.

  - d. increasing atomic number.

# **Periodic Trends Ch 6 Answers**

## **Atef Korchef**

#### **Periodic Trends Ch 6 Answers:**

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