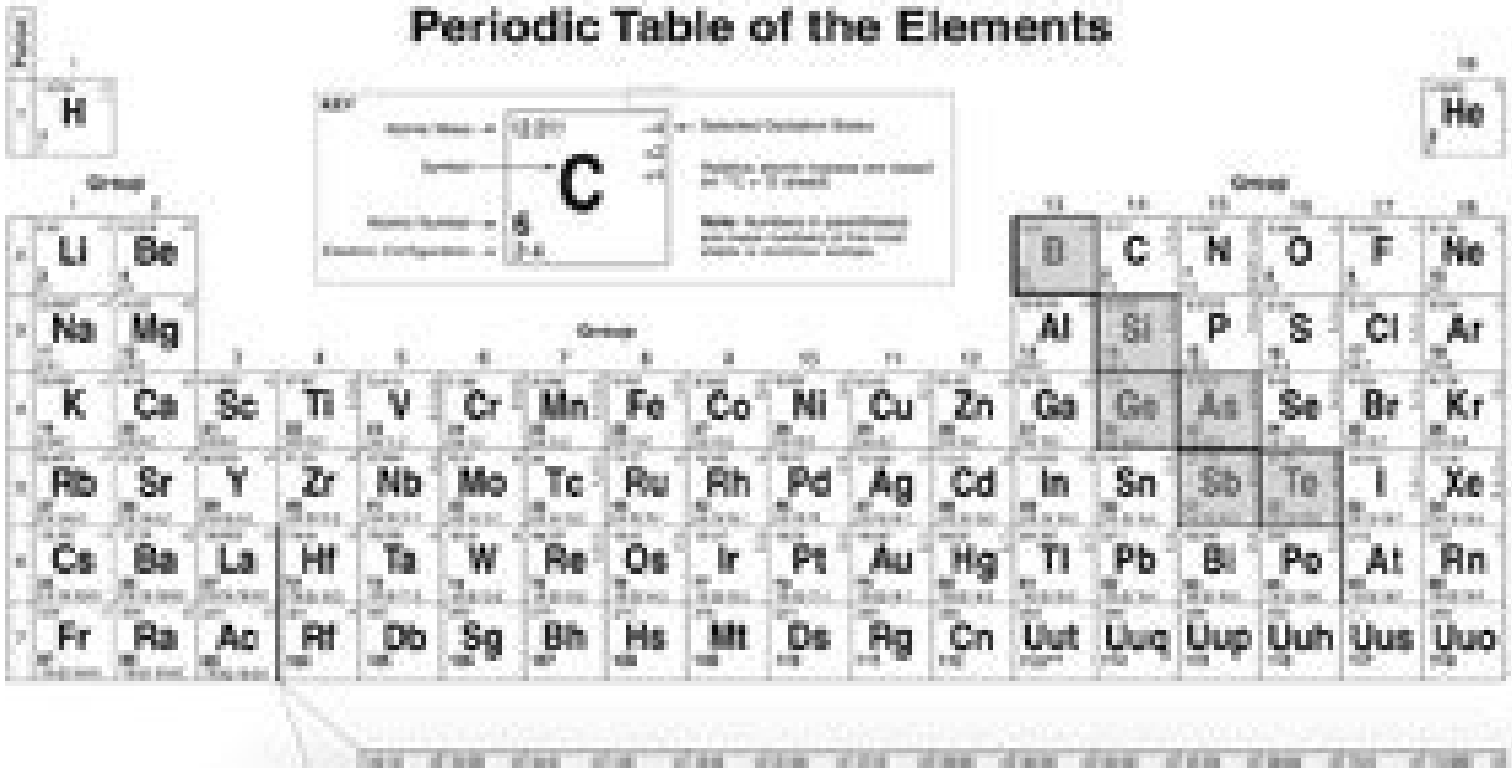
Unit 2: The Periodic Table Review Sheet

U2L1: PT History

- On the present day Periodic Table, elements are arranged in order of increasing atomic number
- **Groups**: vertical columns on the periodic table
  - Elements in the same group have similar chemical properties - because they have the same number of valence electrons
    - Valence electron: outermost electrons (last # in the electron configuration)
- **Periods**: horizontal rows on the periodic table

U2L2: Parts of the Periodic Table

- Elements are classified as either metals, nonmetals, or metalloids
  - Classification can be determined by location on the periodic table or by different properties



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