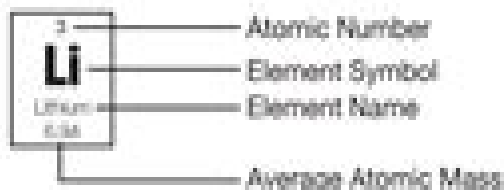


The Periodic Table of the Elements

The Periodic Table of the Elements																							
<div><div>1</div><div>H</div><div>Hydrogen</div><div>1.01</div></div>		<div><div><div>3</div><div>Li</div><div>Lithium</div><div>6.94</div></div><div>Atomic Number</div><div>Element Symbol</div><div>Element Name</div><div>Average Atomic Mass</div></div>																<div><div>2</div><div>He</div><div>Helium</div><div>4.00</div></div>					
<div><div>3</div><div>Li</div><div>Lithium</div><div>6.94</div></div>	<div><div>4</div><div>Be</div><div>Beryllium</div><div>9.01</div></div>																	<div><div>5</div><div>B</div><div>Boron</div><div>10.81</div></div>	<div><div>6</div><div>C</div><div>Carbon</div><div>12.01</div></div>	<div><div>7</div><div>N</div><div>Nitrogen</div><div>14.01</div></div>	<div><div>8</div><div>O</div><div>Oxygen</div><div>16.00</div></div>	<div><div>9</div><div>F</div><div>Fluorine</div><div>18.99</div></div>	<div><div>10</div><div>Ne</div><div>Neon</div><div>20.18</div></div>
<div><div>11</div><div>Na</div><div>Sodium</div><div>22.99</div></div>	<div><div>12</div><div>Mg</div><div>Magnesium</div><div>24.31</div></div>																	<div><div>13</div><div>Al</div><div>Aluminum</div><div>26.98</div></div>	<div><div>14</div><div>Si</div><div>Silicon</div><div>28.09</div></div>	<div><div>15</div><div>P</div><div>Phosphorus</div><div>30.97</div></div>	<div><div>16</div><div>S</div><div>Sulfur</div><div>32.07</div></div>	<div><div>17</div><div>Cl</div><div>Chlorine</div><div>35.45</div></div>	<div><div>18</div><div>Ar</div><div>Argon</div><div>39.95</div></div>
<div><div>19</div><div>K</div><div>Potassium</div><div>39.10</div></div>	<div><div>20</div><div>Ca</div><div>Calcium</div><div>40.08</div></div>	<div><div>21</div><div>Sc</div><div>Scandium</div><div>44.96</div></div>	<div><div>22</div><div>Ti</div><div>Titanium</div><div>47.87</div></div>	<div><div>23</div><div>V</div><div>Vanadium</div><div>50.94</div></div>	<div><div>24</div><div>Cr</div><div>Chromium</div><div>52.00</div></div>	<div><div>25</div><div>Mn</div><div>Manganese</div><div>54.94</div></div>	<div><div>26</div><div>Fe</div><div>Iron</div><div>55.85</div></div>	<div><div>27</div><div>Co</div><div>Cobalt</div><div>58.93</div></div>	<div><div>28</div><div>Ni</div><div>Nickel</div><div>58.69</div></div>	<div><div>29</div><div>Cu</div><div>Copper</div><div>63.55</div></div>	<div><div>30</div><div>Zn</div><div>Zinc</div><div>65.38</div></div>	<div><div>31</div><div>Ga</div><div>Gallium</div><div>69.72</div></div>	<div><div>32</div><div>Ge</div><div>Germanium</div><div>72.61</div></div>	<div><div>33</div><div>As</div><div>Arsenic</div><div>74.92</div></div>	<div><div>34</div><div>Se</div><div>Selenium</div><div>78.96</div></div>	<div><div>35</div><div>Br</div><div>Bromine</div><div>79.90</div></div>	<div><div>36</div><div>Kr</div><div>Krypton</div><div>83.80</div></div>						
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How to Read the Periodic Table of Elements | Chemistry for Beginners Grade 5 | Children's Science & Nature Books Baby Professor,2021-11-01 At the end of this book your child should be able to read and understand some of the periodic table He she should gain the basic understanding of how elements can be identified in symbols and that the table can be read from left to right Moreover your child will also learn what atomic numbers mean and how the table is divided between elements Start reading today **The Elements** Adrian Dingle,2017

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