

## 2.6 - Periodic Trends worksheet

Name \_\_\_\_\_

### I. Short answer

1. Label each group on the periodic table.
2. Explain what valence electrons are.
3. Explain what makes an element stable?

### II. Answer the questions using the symbols on the periodic table

\_\_\_\_\_ 1. What elements are in period 5?

What element(s) is a

\_\_\_\_\_ 2. halogens

\_\_\_\_\_ 3. lanthanides

\_\_\_\_\_ 4. alkaline earth metals

\_\_\_\_\_ 5. noble gases

\_\_\_\_\_ 6. transition metals

\_\_\_\_\_ 7. alkali metals

\_\_\_\_\_ 8. actinides

What elements are in the sublevel

\_\_\_\_\_ 9. s

\_\_\_\_\_ 10. p

\_\_\_\_\_ 11. d

\_\_\_\_\_ 12. f

What elements are

\_\_\_\_\_ 13. metals,

\_\_\_\_\_ 14. nonmetals

\_\_\_\_\_ 15. metalloids,

\_\_\_\_\_ 16. Which element is the most reactive nonmetal?

\_\_\_\_\_ 17. Which element is the most reactive metal?

\_\_\_\_\_ 18. Which element below is the least reactive?

\_\_\_\_\_ 19. Which element has the largest radius in period 4?

\_\_\_\_\_ 20. Which element has the largest radius in the Alkali metals family?

\_\_\_\_\_ 21. Which element has the highest ionization energy in period four?

\_\_\_\_\_ 22. Which elements has the second largest atomic number?

\_\_\_\_\_ 23. Which element has the largest atomic mass?

\_\_\_\_\_ 24. Which element is located at period 4, group 15?

\_\_\_\_\_ 25. Which element is located at period 5, group 2?

\_\_\_\_\_ 27. Which elements will form +1 ions?

\_\_\_\_\_ 28. Which group of metals will have the same properties

\_\_\_\_\_ 29. Which group of nonmetals will have the same properties

\_\_\_\_\_ 30. On the back, define all chapter 5, section 3 vocabulary terms.

b																	r		
																	h		o
	a																		u
i								t											
				e														z	

		w												
					y									



# Periodic Table Chemistry Answers

**O García**



## **Periodic Table Chemistry Answers:**

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*Regents Exams and Answers: Chemistry--Physical Setting Revised Edition* Barron's Educational Series, Albert Tarendash, 2021-01-05 Barron's Regents Exams and Answers Chemistry provides essential practice for students taking the Chemistry Regents including actual recently administered exams and thorough answer explanations for all questions This book features Eight actual administered Regents Chemistry exams so students can get familiar with the test Thorough explanations for all answers Self analysis charts to help identify strengths and weaknesses Test taking techniques and strategies A detailed outline of all major topics tested on this exam A glossary of important terms to know for test day **E3**

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