

Trends on the Periodic Table

Part 1: Identifying chemical properties/characteristics of the elements

For each of the following elements, draw a caricature or cartoon above the description to show its "chemical personality" based on the description provided. (Note: See the Neon example for ideas.)

Lithium
Being one of the smaller elements with only 3 electrons, lithium is happy to give one of its electrons away and never seems to ask for any in return.



Fluorine
One of the smallest elements, Fluorine is often described as snappy with a "magnetic" personality. As part of the halogen family (notorious electron thieves), Fluorine is the best at convincing other elements to give up their electrons to him. However, he holds on tight to his own electrons and will not let them go without a fight.

I am so perfect... I have a full electron shell. I don't need those other elements... I hope they just leave me alone!



Neon
Part of the Noble family, Neon does not readily interact with other elements. With a full set of valence electrons, Neon feels that he has everything he needs to make him happy without sharing with others.

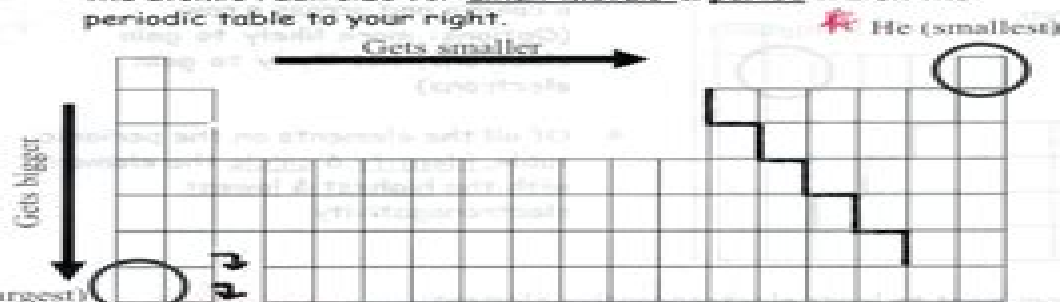
Francium
Francium is the largest of all of the elements and has so many electrons that it has trouble keeping track of them. Francium will eagerly give up electrons to other elements and never takes in extras. But watch out, it has an extremely reactive and explosive personality!

Iodine
As part of the halogen family, Iodine is also a skilled electron thief, however, not nearly as good as its smaller brothers and sisters. Iodine will also put up a pretty good fight to keep its own electrons.

Part 2: Use your textbook to complete the following:

Atomic Radii Size:

1. Explain **atomic radius**: A measure of the size (volume) of an atom. Determined by measuring half of the distance between two identical atoms bonded together.
2. Using the information in figure 5-13 (pg 141) in your book, **write** the atomic radii size for **alkali metals & period #2** on the periodic table to your right.



3. What happens to atomic radii as you move down a column and across a row? On the arrows on the diagram, indicate the trend. (Options: gets bigger OR gets smaller)
4. Of all the elements on the periodic table, **identify & circle the largest & smallest** in size.

5. Rank the following elements from largest to smallest in atomic radii:

a) Be, Ba, Mg = **Ba, Mg, Be**

b) Li, F, B = **Li, B, F**

6. How can you explain the trend?

a. Why do they get smaller as you go across a period? Additional protons in the nucleus "pull" on electrons, bring the electron cloud closer.

Effective nuclear charge, Z_{eff} , increases the attraction of the nucleus and therefore pulls the electron cloud closer to the nucleus resulting in smaller radii.

not a good enough answer
4AP Answer

Periodic Table Answers Pogil

Minjie Lin



Periodic Table Answers Pogil:

Process Oriented Guided Inquiry Learning (POGIL) Richard Samuel Moog, 2008 POGIL is a student centered group learning pedagogy based on current learning theory This volume describes POGIL s theoretical basis its implementations in diverse environments and evaluation of student outcomes

Transforming Urban Education Kenneth Tobin, Ashraf Shady, 2014-04-03 Transformations in Urban Education Urban Teachers and Students Working Collaboratively addresses pressing problems in urban education contextualized in research in New York City and nearby school districts on the Northeast Coast of the United States The schools and institutions involved in empirical studies range from elementary through college and include public and private schools alternative schools for dropouts and museums Difference is regarded as a resource for learning and equity issues are examined in terms of race ethnicity language proficiency designation as special education and gender The contexts for research on teaching and learning involve science mathematics uses of technology literacy and writing comic books A dual focus addresses research on teaching and learning and learning to teach in urban schools Collaborative activities addressed explicitly are teachers and students enacting roles of researchers in their own classrooms cogenative dialogues as activities to allow teachers and students to learn about one another s cultures and express their perspectives on their experienced realities and negotiate shared recommendations for changes to enacted curricula Coteaching is also examined as a means of learning to teach teaching and learning and undertaking research The scholarship presented in the constituent chapters is diverse reflecting multi logicity within sociocultural frameworks that include cultural sociology cultural historical activity theory prosody sense of place and hermeneutic phenomenology Methodologies employed in the research include narratology interpretive reflexive and authentic inquiry and multi level inquiries of video resources combined with interpretive analyses of social artifacts selected from learning environments This edited volume provides insights into research of places in which social life is enacted as if there were no research being undertaken The research was intended to improve practice Teachers and learners as research participants were primarily concerned with teaching and learning and as a consequence as we learned from research participants were made aware of what we learned the purpose being to improve learning environments Accordingly research designs are contingent on what happens and emergent in that what we learned changed what happened and expanded possibilities to research and learn about transformation through heightening participants awareness about possibilities for change and developing interventions to improve learning

Cooperative Learning in Higher Education Barbara Millis, 2023-07-03 Research has identified cooperative learning as one of the ten High Impact Practices that improve student learning If you ve been interested in cooperative learning but wondered how it would work in your discipline this book provides the necessary theory and a wide range of concrete examples Experienced users of cooperative learning demonstrate how they use it in settings as varied as a developmental mathematics course at a community college and graduate courses in history and the sciences and how it

works in small and large classes as well as in hybrid and online environments The authors describe the application of cooperative learning in biology economics educational psychology financial accounting general chemistry and literature at remedial introductory and graduate levels The chapters showcase cooperative learning in action at the same time introducing the reader to major principles such as individual accountability positive interdependence heterogeneous teams group processing and social or leadership skills The authors build upon and cross reference each others chapters describing particular methods and activities in detail They explain how and why they may differ about specific practices while exemplifying reflective approaches to teaching that never fail to address important assessment issues **Choice** ,2009

Atoms, Molecules & Elements: Patterns In the Periodic Table Gr. 5-8 George Graybill,2015-10-01 This is the chapter slice Patterns In the Periodic Table from the full lesson plan Atoms Molecules Elements Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Students will label each part of the atom learn what compounds are and explore the patterns in the periodic table of elements to find calcium Ca chlorine Cl and helium He through hands on activities These and more science concepts are presented in a way that makes them more accessible to students and easier to understand Written to grade and using simplified language and vocabulary and comprised of reading passages student activities crossword word search comprehension quiz and color mini posters our resource can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom s Taxonomy and STEM initiatives *Atoms, Molecules & Elements: The Periodic Table Gr. 5-8* George Graybill,2015-10-01 This is the

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periodic table through text written at four different reading levels Symbols on the pages represent reading level ranges to help differentiate instruction Provided comprehension questions complement the text *Atoms, Molecules & Elements: What Are Elements? Gr. 5-8* George Graybill,2015-10-01 This is the chapter slice What Are Elements from the full lesson plan Atoms Molecules Elements Young scientists will be thrilled to explore the invisible world of atoms molecules and elements

Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary. Students will label each part of the atom, learn what compounds are, and explore the patterns in the periodic table of elements to find calcium (Ca), chlorine (Cl), and helium (He) through hands-on activities. These and more science concepts are presented in a way that makes them more accessible to students and easier to understand. Written to grade and using simplified language and vocabulary, and comprised of reading passages, student activities, crossword, word search, comprehension quiz, and color mini posters, our resource can be used effectively for test prep and your whole class. All of our content is aligned to your State Standards and are written to Bloom's Taxonomy and STEM initiatives.

Atoms, Molecules & Elements Gr. 5-8 George Graybill, 2007-09-01 Young scientists will be thrilled to explore the invisible world of atoms, molecules, and elements. Our resource makes the periodic table easier to understand. Begin by answering what are atoms. See how the atomic model is made up of electrons, protons, and neutrons. Find out what a molecule is and how they differ from elements. Then move on to compounds. Find the elements that make up different compounds. Get comfortable with the periodic table by recognizing each element as part of a group. Examine how patterns in the periodic table dictate how those elements react with others. Finally, explore the three important kinds of elements: metals, nonmetals, and inert gases. Aligned to the Next Generation Science Standards and written to Bloom's Taxonomy and STEAM initiatives, additional hands-on experiments, crossword, word search, comprehension quiz, and answer key are also included.

The Basics of the Periodic Table Leon Gray, 2013-12-15 Provides basic information on the periodic table. Includes biographical information on Dmitri Mendeleev, color photographs and diagrams, sidebars, a glossary, and further reading sources.

Elements and the Periodic Table, Grades 5 - 8 Abby, 2013-01-02 Aligned to Common Core State Standards, Elements and the Periodic Table present the basics of the Periodic Table in an easy to understand, easy to master way. It contains fun activities, transparency masters, quizzes, tests, rubrics, grading sheets, and more. From basic elements to table organization, Elements and the Periodic Table is the essential handbook for middle school science.

The HISTORY of the ATOM, the PERIODIC TABLE and RADIOACTIVITY Alex Kimmons, 2021-05-15 THE HISTORY OF THE ATOM, THE PERIODIC TABLE and RADIOACTIVITY monochrome. Chemistry is an interesting and fundamental branch of science because it gives us the chance to explain the secrets of nature. What is water? What do we use in our cars as fuel? What is aspirin? What are perfumes made of? These kinds of questions and their answers are all part of the world of chemistry. There is no industry that does not depend upon chemical substances: the petroleum, pharmaceuticals, garment, aircraft, steel, and electronics industries, for example, as well as agriculture, all utilize the science of chemistry. This book helps everyone to understand nature. However, one does not need to be a chemist or scientist to understand the simplicity within the complexity around us. The aim was to write a modern, up-to-date book where students and teachers can get concise information about the structure of substances. Sometimes reactions are given in the detailed form, but in general, excessive detail has been omitted. The book is designed to introduce fundamental knowledge in three

areas the history of the atom the periodic table and radioactivity We will study the historical development of atomic structure theories the tendencies of elements in periods and groups and the types of emissions and uses of radioactivity *Elements and the Periodic Table, Grades 5 - 12* Theodore S. Abbgy, 2013-01-02 Aligned to Common Core State Standards Elements and the Periodic Table present the basics of the Periodic Table in an easy to understand easy to master way It contains fun activities transparency masters quizzes tests rubrics grading sheets and more From basic elements to table organization Elements and the Periodic Table is the essential handbook for middle school science *Mastering the Periodic Table* Linda Trombley, Faye Williams, 2000 Whether students are studying chemistry biology or other sciences the periodic table is a vitally important tool These 50 word games puzzles and other creative activities unlock the nature of the various elements while explicating periodicity atomic structure element groups and more Complete teacher support includes background information answer keys and materials lists Atoms, Molecules & Elements: What Are Compounds? Gr. 5-8 George Graybill, 2015-10-01 This is the chapter slice What Are Compounds from the full lesson plan Atoms Molecules Elements Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Students will label each part of the atom learn what compounds are and explore the patterns in the periodic table of elements to find calcium Ca chlorine Cl and helium He through hands on activities These and more science concepts are presented in a way that makes them more accessible to students and easier to understand Written to grade and using simplified language and vocabulary and comprised of reading passages student activities crossword word search comprehension quiz and color mini posters our resource can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom s Taxonomy and STEM initiatives Mystery of the Periodic Table Benjamin D Wiker, 2003-04-18 Leads the reader on a delightful and absorbing journey through the ages on the trail of the elements of the Periodic Table as we know them today He introduces the young reader to people like Von Helmont Boyle Stahl Priestly Cavendish Lavoisier and many others all incredibly diverse in personality and approach who have laid the groundwork for a search that is still unfolding to this day The first part of Wiker s witty and solidly instructive presentation is most suitable to middle school age while the later chapters are designed for ages 12 13 and up with a final chapter somewhat more advanced Illustrated by Jeanne Bendick and Ted Schluenderfritz **Elemental** Tim James, 2019-03-26 If you want to understand how our world works the periodic table holds the answers When the seventh row of the periodic table of elements was completed in June 2016 with the addition of four final elements nihonium moscovium tennessine and oganesson we at last could identify all the ingredients necessary to construct our world In Elemental chemist and science educator Tim James provides an informative entertaining and quirkily illustrated guide to the table that shows clearly how this abstract and seemingly jumbled graphic is relevant to our day to day lives James tells the story of the periodic table from its ancient Greek roots when you could count

the number of elements humans were aware of on one hand to the modern alchemists of the twentieth and twenty first centuries who have used nuclear chemistry and physics to generate new elements and complete the periodic table In addition to this he answers questions such as What is the chemical symbol for a human What would happen if all of the elements were mixed together Which liquid can teleport through walls Why is the medieval dream of transmuting lead into gold now a reality Whether you re studying the periodic table for the first time or are simply interested in the fundamental building blocks of the universe from the core of the sun to the networks in your brain Elemental is the perfect guide *Science Spectrum* Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2004

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The Periodic Table Mary Griffin,2018-12-15 The periodic table is the way scientists have organized the known elements but up and coming scientists can learn much from the periodic table as well Your readers will learn how to read the periodic table including what an element s atomic number means and what the rows and columns signify They ll gather facts about common and interesting elements and even discover how they can predict how elements might chemically react from their position in the table Fact boxes and images add additional information to the accessible text

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Table of Contents Periodic Table Answers Pogil

1. Understanding the eBook Periodic Table Answers Pogil
 - The Rise of Digital Reading Periodic Table Answers Pogil
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Answers Pogil
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Table Answers Pogil
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Answers Pogil
 - Personalized Recommendations

- Periodic Table Answers Pogil User Reviews and Ratings
- Periodic Table Answers Pogil and Bestseller Lists
- 5. Accessing Periodic Table Answers Pogil Free and Paid eBooks
 - Periodic Table Answers Pogil Public Domain eBooks
 - Periodic Table Answers Pogil eBook Subscription Services
 - Periodic Table Answers Pogil Budget-Friendly Options
- 6. Navigating Periodic Table Answers Pogil eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Table Answers Pogil Compatibility with Devices
 - Periodic Table Answers Pogil Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Table Answers Pogil
 - Highlighting and Note-Taking Periodic Table Answers Pogil
 - Interactive Elements Periodic Table Answers Pogil
- 8. Staying Engaged with Periodic Table Answers Pogil
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Table Answers Pogil
- 9. Balancing eBooks and Physical Books Periodic Table Answers Pogil
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Table Answers Pogil
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Table Answers Pogil
 - Setting Reading Goals Periodic Table Answers Pogil
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Table Answers Pogil
 - Fact-Checking eBook Content of Periodic Table Answers Pogil

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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