

Vicinal dibromide formation:

2. In solvent like H.O.



Halohydrin formation

- + It is a regionelective reaction.
- The nucleophilic attack of the water occurs at the more substituted carbon atom.

2. With Alkymes

- · It is an anti-addition reaction.
- 1. In solvent like CCI, CH,CI, etc.



Tetrabormide formation

OH

3. With Aromatic Compounds



4. Photo halogenation of alkanes

5. Alpha Bromination of carbonyl compounds

6. Oxidation of Monosaccharides

- Bromine water is a mild oxidizing agent.
- selectively exidizes the aldehyde group (for example CHO group of glucose) to a carboxylic acid.

HO

OH

OH

7. Haloform Reaction

 converts methyl ketones to Haloform (CHX₃(X=Cl,Br,I)) and salt of carboxylic acids.

8. The Hoffmann Bromamide degradation

 It converts primary amide to a primary amine with one less carbon atom.

9. Hell-Volhard-Zelinski (or HVZ) reaction

5. CN

Cyanide Ion and Nitrile Group

1. Conversion of Nitriles to a Primary Amine

2. Conversion of Nitriles to Aldehydes

3. Reaction of Nitriles with Grignard reagent

4. Hydrolysis of Nitriles to Carboxylic Acids

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Table of Contents Ochem Reagent Guide

- 1. Understanding the eBook Ochem Reagent Guide
 - The Rise of Digital Reading Ochem Reagent Guide
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Ochem Reagent Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ochem Reagent Guide
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Ochem Reagent Guide
 - Personalized Recommendations
 - Ochem Reagent Guide User Reviews and Ratings
 - Ochem Reagent Guide and Bestseller Lists

- 5. Accessing Ochem Reagent Guide Free and Paid eBooks
 - Ochem Reagent Guide Public Domain eBooks
 - Ochem Reagent Guide eBook Subscription Services
 - Ochem Reagent Guide Budget-Friendly Options
- 6. Navigating Ochem Reagent Guide eBook Formats
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 - \circ Benefits of a Digital Library
 - o Creating a Diverse Reading Collection Ochem Reagent Guide
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Ochem Reagent Guide
 - Setting Reading Goals Ochem Reagent Guide
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Ochem Reagent Guide
 - Fact-Checking eBook Content of Ochem Reagent Guide
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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