

Questions with Solution P. K. Nag

Q4.1 An engine is tested by means of a water brake at 1000 rpm. The measured torque of the engine is 10000 mN and the water consumption of the brake is 0.5 m³/s, its inlet temperature being 20°C. Calculate the water temperature at exit, assuming that the whole of the engine power is ultimately transformed into heat which is absorbed by the cooling water.

(Ans. 20.5°C)

Solution: Power = $T\omega$
 $= 10000 \times \left(\frac{2\pi \times 1000}{60} \right)$
 $= 1.0472 \times 10^6 \text{ W}$
 $= 1.0472 \text{ MW}$

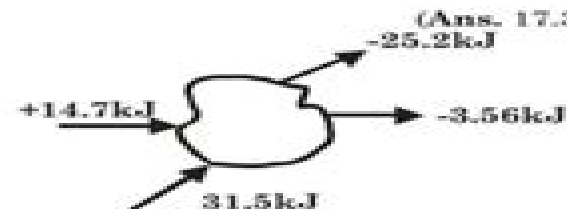
Let final temperature = $t^\circ\text{C}$

$$\begin{aligned}\therefore \text{Heat absorb by cooling water / unit} &= \dot{m} \times \Delta t \\ &= \dot{V} \rho \times \Delta t \\ &= 0.5 \times 1000 \times 4.2 \times (t - 20) \\ \therefore 0.5 \times 1000 \times 4.2 \times (t - 20) &= 1.0472 \times 10^6 \\ \therefore t - 20 &= 0.4986 \approx 0.5 \\ \therefore t &= 20.5^\circ\text{C}\end{aligned}$$

Q4.2 In a cyclic process, heat transfers are + 14.7 kJ, - 25.2 kJ, - 3.56 kJ and + 31.5 kJ. What is the net work for this cyclic process?

(Ans. 17.34 kJ)

Solution : $\sum Q = (14.7 + 31.5 - 25.2 - 3.56) \text{ kJ}$
 $= 17.44 \text{ kJ}$
 From first law of thermodynamics
 (for a cyclic process)
 $\sum Q = \sum W$
 $\therefore \sum W = 17.44 \text{ kJ}$



Q4.3 A slow chemical reaction takes place in a fluid at the constant pressure of 0.1 MPa. The fluid is surrounded by a perfect heat insulator during the reaction which begins at state 1 and ends at state 2. The insulation is then removed and 105 kJ of heat flow to the surroundings as the fluid goes to state 3. The following data are observed for the fluid at states 1, 2 and 3.

State	$v \text{ (m}^3\text{)}$	$t \text{ (}^\circ\text{C)}$
1	0.003	20
2	0.3	370
3	0.06	20

For the fluid system, calculate E_2 and E_3 , if $E_1 = 0$

(Ans. $E_2 = -29.7 \text{ kJ}$, $E_3 = -110.7 \text{ kJ}$)

Pk Nag Solution Manual

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- Chapter 1: Introduction to Pk Nag Solution Manual
- Chapter 2: Essential Elements of Pk Nag Solution Manual
- Chapter 3: Pk Nag Solution Manual in Everyday Life
- Chapter 4: Pk Nag Solution Manual in Specific Contexts
- Chapter 5: Conclusion

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Table of Contents Pk Nag Solution Manual

1. Understanding the eBook Pk Nag Solution Manual

- The Rise of Digital Reading Pk Nag Solution Manual
- Advantages of eBooks Over Traditional Books
- 2. Identifying Pk Nag Solution Manual
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Pk Nag Solution Manual
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Pk Nag Solution Manual
 - Personalized Recommendations
 - Pk Nag Solution Manual User Reviews and Ratings
 - Pk Nag Solution Manual and Bestseller Lists
- 5. Accessing Pk Nag Solution Manual Free and Paid eBooks
 - Pk Nag Solution Manual Public Domain eBooks
 - Pk Nag Solution Manual eBook Subscription Services
 - Pk Nag Solution Manual Budget-Friendly Options
- 6. Navigating Pk Nag Solution Manual eBook Formats
 - ePub, PDF, MOBI, and More
 - Pk Nag Solution Manual Compatibility with Devices
 - Pk Nag Solution Manual Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Pk Nag Solution Manual
 - Highlighting and Note-Taking Pk Nag Solution Manual
 - Interactive Elements Pk Nag Solution Manual
- 8. Staying Engaged with Pk Nag Solution Manual
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Pk Nag Solution Manual

9. Balancing eBooks and Physical Books Pk Nag Solution Manual
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Pk Nag Solution Manual
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Pk Nag Solution Manual
 - Setting Reading Goals Pk Nag Solution Manual
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Pk Nag Solution Manual
 - Fact-Checking eBook Content of Pk Nag Solution Manual
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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