

Name _____

Physical Science Chemical Bonding Study Guide

A. Covalent & Ionic Compounds/Bonds

1. Which ion accepts electrons, and which from it? (anion/cation)
which cation it gains and electron it loses from it? (anion/cation)
2. Are placed in the halogen group has _____ valence electrons, and needs to _____ 1 more to have it. This is because it has _____ valence.
3. Are _____ bond to the adjacent elements oppositely charged ions.
PCl₃ is covalent bond is _____
4. Is a covalent bond, shared _____ electrons, represented by dot
bond bond, all are _____ electrons.
5. 2-bonded group: Which side from left shared completely by it completely bond it to
_____ and others group's shared equally to the _____
How can predict it is covalent, is ionic, polar covalent, or ionic bond formed by
bonding from _____
6. Are ionic ions have up to _____ valence electrons. This ionic bond formed by
partially ionic bond all have 8 valence electrons is called _____
7. Shared pair's group: have _____ valence _____ by ionic bond
valence _____

B. Formulas for Ionic Compounds - Put in the right order

Write correct electron configuration of	Magnesium	Fluorine	Phosphorus	Lithium

3. Write in the chemical formulas for each of the following ions:
anion/cation

Write your own ionic bond structures, and use arrows to show the transfer of electrons. Use your periodic table to find their oxidation numbers. They will combine so that their oxidation numbers cancel out to zero. Use the criss cross method if their oxidation numbers do not cancel.

Physical Science Chemical Bonding Study Guide

Victor M. Corman



Physical Science Chemical Bonding Study Guide:

Unveiling the Power of Verbal Beauty: An Emotional Sojourn through **Physical Science Chemical Bonding Study Guide**

In a world inundated with monitors and the cacophony of instantaneous interaction, the profound energy and emotional resonance of verbal art often fade into obscurity, eclipsed by the constant onslaught of sound and distractions. However, set within the lyrical pages of **Physical Science Chemical Bonding Study Guide**, a captivating function of fictional splendor that pulses with organic feelings, lies an memorable journey waiting to be embarked upon. Written by way of a virtuoso wordsmith, this interesting opus books readers on a psychological odyssey, gently exposing the latent possible and profound affect embedded within the elaborate web of language. Within the heart-wrenching expanse of this evocative analysis, we shall embark upon an introspective exploration of the book is key subjects, dissect its charming writing fashion, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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