

Fluid and Electrolytes: Potter and Perry Chapter 42

- Overview
 - Fluid, electrolyte, and acid-base balances within the body maintain health and function of all body systems
 - Fluid: water that contains dissolved or suspended substances such as glucose, mineral salts, and proteins
 - Characteristics of body fluids influence body system function and have regulatory mechanisms that keep them in balance:
 - Fluid amount → volume
 - Concentration → osmolality
 - Composition → electrolyte concentration
 - Degree of acidity → pH
 - Water = 60% of adult body weight
 - Decreases with age
 - Fluid distribution: movement of fluid among its various compartments
 - Fluid located in 2 compartments:
 - Extracellular (ECF): outside of the cell
 - Intravascular – liquid part of the blood (plasma)
 - Interstitial – located between the cells and outside the blood vessels (within the tissues)
 - Intracellular (ICF): inside of the cell
 - Body tries to maintain homeostasis
 - Osmolality: measure of the # of particles per kilogram of water
 - Normal range: 285 – 295 mOsm/kg H₂O
 - Tonicity: concentration of solutes in solution
 - Isotonic – same tonicity as normal blood (no fluid shift)
 - Hypotonic – more dilute than the blood
 - Hypertonic – more concentration than normal blood
 - Electrolytes: a compound that separates into ions when it dissolves in water
 - Cations: + charged
 - Na, K, Ca, Mg
 - Anions: negatively charged
 - Cl, HCO₃
- Movement of Water and Electrolytes:
 - Active transport: requires energy (ATP) to move electrolytes across cell membranes against concentration gradient (areas of low concentration to areas of high concentration)
 - Example: Sodium potassium pump
 - Moves Na out of a cell and K into it, keeping ICF lower in Na and higher in K than the ECF

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