



MERCY JUNIOR COLLEGE
Grade 11
Physical Science
Finals Exam

NAME: _____

YEAR LEVEL & SECTION: _____

TEACHER'S NAME: _____

PROCTOR'S NAME: _____

1. MULTIPLE CHOICE

INSTRUCTION: Encircle the letter of the correct answer. (1 POINT EACH)

- It is a subatomic particle, symbol e^- or β^- , with a negative elementary electric charge.
 - Neutron
 - Electron
 - Proton
 - Atom
- It is a subatomic particle, symbol p or p^+ , with a positive electric charge of $+1e$ elementary charge and mass slightly less than that of a neutron.
 - Neutron
 - Electron
 - Proton
 - Atom
- It is a subatomic particle, symbol n or n^0 , with no net electric charge and a mass slightly larger than that of a proton.
 - Neutron
 - Electron
 - Proton
 - Atom
- Is this an example of ionic bonding?



carbon dioxide

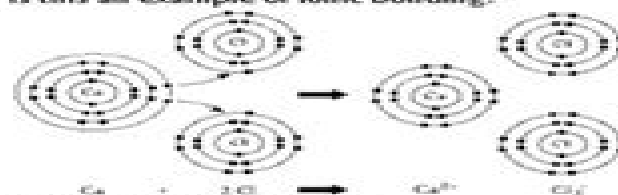
- Yes, it is ionic bonding.
- No, it is of another type of chemical bonding.
- There is no chemical bonding present.
- This is just an orbital model.

- Is this an example of ionic bonding?



lithium fluoride

- Yes, it is ionic bonding.
 - No, it is of another type of chemical bonding.
 - There is no chemical bonding present.
 - This is just an orbital model.
- In atomic physics and quantum chemistry, the electron configuration is the distribution of electrons of an atom or molecule in atomic or molecular orbitals.
 - Electron Configuration
 - Periodic Table of Elements
 - Orbital Models
 - Energy Levels
 - Is this an example of ionic bonding?



- Yes, it is ionic bonding.
- No, it is of another type of chemical bonding.
- There is no chemical bonding present.
- This is just an orbital model.

- It is a chemical rule of thumb that reflects observation that atoms of main-group elements tend to combine in such a way that each atom has eight electrons in its valence shell, giving it the same electronic configuration as a noble gas.
 - Octet Rule
 - Valence Electrons
 - Atomic Orbitals
 - Chemical Bonds
- It is the smallest constituent unit of ordinary matter that has the properties of a chemical element.
 - Neutron
 - Electron
 - Proton
 - Atom
- It is a type of chemical bond that involves the electrostatic attraction between oppositely

Physical Science Grade 11 Final Examination 2014

Scope

S Baum



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Table of Contents Physical Science Grade 11 Final Examination 2014 Scope

1. Understanding the eBook Physical Science Grade 11 Final Examination 2014 Scope
 - The Rise of Digital Reading Physical Science Grade 11 Final Examination 2014 Scope
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Grade 11 Final Examination 2014 Scope
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Grade 11 Final Examination 2014 Scope
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Grade 11 Final Examination 2014 Scope

- Personalized Recommendations
- Physical Science Grade 11 Final Examination 2014 Scope User Reviews and Ratings
- Physical Science Grade 11 Final Examination 2014 Scope and Bestseller Lists
- 5. Accessing Physical Science Grade 11 Final Examination 2014 Scope Free and Paid eBooks
 - Physical Science Grade 11 Final Examination 2014 Scope Public Domain eBooks
 - Physical Science Grade 11 Final Examination 2014 Scope eBook Subscription Services
 - Physical Science Grade 11 Final Examination 2014 Scope Budget-Friendly Options
- 6. Navigating Physical Science Grade 11 Final Examination 2014 Scope eBook Formats
 - ePub, PDF, MOBI, and More
 - Physical Science Grade 11 Final Examination 2014 Scope Compatibility with Devices
 - Physical Science Grade 11 Final Examination 2014 Scope Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science Grade 11 Final Examination 2014 Scope
 - Highlighting and Note-Taking Physical Science Grade 11 Final Examination 2014 Scope
 - Interactive Elements Physical Science Grade 11 Final Examination 2014 Scope
- 8. Staying Engaged with Physical Science Grade 11 Final Examination 2014 Scope
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Grade 11 Final Examination 2014 Scope
- 9. Balancing eBooks and Physical Books Physical Science Grade 11 Final Examination 2014 Scope
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Grade 11 Final Examination 2014 Scope
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Physical Science Grade 11 Final Examination 2014 Scope
 - Setting Reading Goals Physical Science Grade 11 Final Examination 2014 Scope
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physical Science Grade 11 Final Examination 2014 Scope

- Fact-Checking eBook Content of Physical Science Grade 11 Final Examination 2014 Scope
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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