

Physical Science – Chapter 2 Practice Test Answer Key

- _____ **a** 1) Which of the following is a physical property of an element?
a) color b) ability to rust c) flammability d) ability to tarnish
- _____ **c** 2) Which of the following is NOT a characteristic of a compound?
a) has different properties from the elements that formed it
b) pure substance made of two or more elements
c) different samples have different properties
d) can be represented by a chemical formula
- _____ **a** 3) A measurement of the amount of matter in an object is its _____.
a) mass b) density c) volume length
- _____ **c** 4) Every chemical or physical change in matter includes a change in _____.
a) volume b) temperature c) energy d) mass
- _____ **d** 5) A(n) _____ is an example of a homogeneous mixture that is very evenly mixed.
a) compound b) element c) pure substance d) solution
- _____ **c** 6) A chemical bond is _____.
a) a group of atoms that are joined together b) The basic particle of matter
c) The force that holds two atoms together
d) A substance formed from the chemical combination of two or more atoms
- _____ **d** 7) Which of the following is an example of a chemical change?
a) melting butter b) mixing milk and chocolate syrup
c) breaking glass d) burning fuel
- _____ **a** 8) Which of the following is NOT true of atoms?
a) They are composed of molecules. b) They are combined with other atoms.
c) They make up elements. d) They are extremely small.
- _____ **d** 9) A(n) _____ of a pure substance can be observed without changing the substance.
a) chemical property b) atom c) element d) physical property
- _____ **a** 10) _____ is stored in the bonds that hold atoms together.
a) Chemical energy b) Physical energy
c) Thermal energy d) Electromagnetic energy
- _____ **b** 11) A(n) _____ is a group of atoms that are joined together by chemical bonds.
a) roundtuit b) molecule c) element d) substance
- _____ **c** 12) A pure substance made of two or more elements that are chemically combined is a(n) _____.
a) substance b) element c) compound d) atom
- _____ **d** 13) A measure of the average energy of random motion of particles of matter is _____.
a) energy b) running c) eventful d) temperature

Physical Science Chapter 1test Answer Key Id A

Brendan G. Carr



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women writing about their firsthand experiences in today's STEM world

MEASUREMENT, EVALUATION AND ASSESSMENT IN EDUCATION RADHA MOHAN, 2016-06-10 Assessment and evaluation are central to the educational system of a country as they impact the national policy of education besides helping in framing future strategies for growth of the nation. The process of measurement, evaluation and assessment is interlinked with curriculum and teaching-learning methods enabling learning outcomes to be determined. Good quality assessment is instrumental in determining a student's future goal and career path besides impacting all stakeholders of the school system. An effective school system cannot be built where the students are examined on the basis of mere paper and pencil tests. Today schools have access to the use of a range of techniques for student's assessment, teacher performance, school's effectiveness and the evaluation of the school system as a whole. The aim of the book is to provide an in-depth knowledge and understanding of measurement, evaluation and statistics in education both from theoretical and practical aspects. It also offers a concise step-by-step guide that helps in making assessment simple and economical in terms of money and time besides being a boon for any institution. The book contains effective strategies to facilitate enhanced learning by explaining the interpretation of test scores. It has several illustrative examples drawn from the author's experience. Portfolios are proving to be a great way of assessment and how to incorporate them in the teaching-learning process is dealt with in detail. Instructions for constructing rubrics for evaluation of different domains of learning, i.e. cognitive, psychomotor and affective, have also been included in the book. Item analysis, basic statistics and different types of tests are discussed elaborately.

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