

TYBSC SEM VI PHYSICAL CHEMISTRY (MODEL MCQ's)

| | |
|---|---|
| 1 | Which of the following is a uni-univalent electrolyte? a) KCl b) CuCl_2 c) AlCl_3 d) MgCl_2 |
| 2 | The ionic strength of a solution of molality m_i and valency of the ion Z_i is ____ a) $\mu = \frac{1}{2} m_i Z_i^2$ b) $\mu = \sum m_i Z_i^2$ c) $\mu = \frac{1}{2} \sum Z_i^2$ d) $\mu = \frac{1}{2} \sum m_i Z_i^2$ |
| 3 | Which of the following is an example of electrolyte concentration cell with transference? a) $\text{Cu} \text{CuSO}_4 \text{AgNO}_3 \text{Ag}$ b) $\text{Pt}, \text{H}_{2(g)} \text{HCl}_{(aq)} \text{AgCl}_{(s)} \text{Ag}$ c) $\text{Pt}, \text{H}_{2(g)} \text{HCl}_{(aq)}(a_1) \text{AgNO}_3(a_2) \text{Ag}$ d) $\text{Ag} - \text{AgCl} \text{NaCl}(a_1) \text{NaCl}(a_2) \text{AgCl} \cdot \text{Ag}$ |
| 4 | A concentration cell which has salt bridge in its representation is ____ a) electrolyte concentration cell with transference b) Electrode concentration cell c) Electrolyte concentration cell without transference d) Gas concentration cell |
| 5 | The total emf of the electrolyte concentration cell with transference containing the liquid junction potential is given by..... a) $E_j = E_{\text{oxidation}} + E_{\text{reduction}}$ b) $E_{\text{cell}} = E_{\text{oxidation}} + E_{\text{reduction}} + E_j$ c) $E_{\text{cell}} = E_{\text{oxidation}} - E_{\text{reduction}} - E_j$ d) $E_j = E_{\text{oxidation}} - E_{\text{reduction}}$ |
| 6 | The activity 'a' is related to the activity coefficient ' γ ' by the equation____ a) $a = \frac{m}{\gamma}$ b) $\gamma = a - m$ c) $a = m + \gamma$ d) $a = m\gamma$ |

Physical Chemistry Multiple Choice Questions

Rachel S Tattersall



Physical Chemistry Multiple Choice Questions:

Multiple Choice Questions in Physical Chemistry for Undergraduate Students Dr Kusum Baser, 2022-08-25

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Seiffert, 2023-01-30 This book introduces the concepts of physical chemistry of polymers in a format targeted for a blended learning approach It provides a basis to bridge polymer chemistry which targets microscopic chain structures and polymer engineering which targets macroscopic material properties and functions Topics covered are single chain statistics multi chain interactions and chain dynamics both from a viewpoint of structure properties mostly mechanical ones and their interrelation In all that the author encourages the reader to think conceptually Explains complex facts through simplifying models diagrams and illustrations Accessible to chemists chemical engineers materials scientists and physicists Tailored content for an interactive blended learning format *Cracking the GRE Chemistry Test*, 2002 The Princeton Review realizes

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