

Use the partial periodic table to answer questions 4 -7.

6
C
Carbon
12.0107

— Atomic number
 — Symbol
 — Name
 — Average Atomic Mass

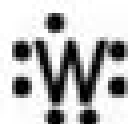
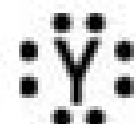
Partial Periodic Table of the Elements

	IA 1		IIA 2		VIA 16	VIIA 17	VIII 18
1	H Hydrogen 1.00794		He Helium 4.0026				
2	Li Lithium 6.941		Be Beryllium 9.0122				
3	Na Sodium 22.989769		Mg Magnesium 24.304				
4	K Potassium 39.0983		Ca Calcium 40.078				

	IIIA 13	IVA 14	VA 15	VIA 16	VIIA 17	VIII 18
	B Boron 10.811	C Carbon 12.0107	N Nitrogen 14.0064	O Oxygen 15.9994	F Fluorine 18.9984	Ne Neon 20.1797
	Al Aluminum 26.981538	Si Silicon 28.0855	P Phosphorus 30.97376	S Sulfur 32.06	Cl Chlorine 35.453	Ar Argon 39.948

Question 5

Suppose scientists discovered four new elements (W, X, Y, Z) while studying rock and soil samples brought back from a Mars mission. Which Lewis dot structure represents an element that should be placed in column VIIA (17) of the periodic table?

- A.  B.  C.  D. 

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J Ma



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