

Part B. Concept Review

Directions: *Correctly complete each sentence by underlining the best of the three choices in parentheses.*

1. (Physical changes, Density, Chemical changes) can be used to identify unknown compounds and elements.
2. The states of matter are (acid, liquid, base), solid, gas, and plasma.
3. Density is a (size-dependent, size-independent, variable) property.
4. Acids have a pH (between 0 and 7, exactly 7, between 7 and 14).
5. Bases have a pH (between 0 and 7, exactly 7, between 7 and 14).
6. You should (always, sometimes, never) taste or smell anything in the lab.
7. Crushing a leaf causes a (physical property, physical change, chemical change).
8. A burning candle is an example of a (physical change, chemical property, chemical change).
9. A sign that a (physical change, chemical change, chemical property) has occurred is the release or absorption of energy.
10. When rocks are crushed, a (physical property, physical change, chemical change) has occurred.



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Dennis G. Zill, Warren S. Wright



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