

Harvard Reference examples

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British Standards Institute. (1990). BS5605:1990. Recommendations for citing and referencing published material. Milton Keynes: BSI.
Neville, C. (2007). The Complete Guide to Referencing and Avoiding Plagiarism. Open UP Study Skills. Maidenhead: Open University Press.
Pears, R. & Shields, G. (2006) Cite Them Right: The Essential Guide to Referencing and Plagiarism. Newcastle-upon-Tyne: Northumbria University Press.

Acts of Parliament (UK Statutes)

Format:

Country. Name of Act: Name of Sovereign. Chapter Number – in italics or underlined. (Year of Publication – in brackets). Place of Publication: Publisher.

Bibliography example:

Great Britain. Human Rights Act 1998: Elizabeth II. Chapter 42. (1998). London: The Stationery Office

In-text example:

The Human Rights Act 1998 indicated that.....

"Direct quotations are placed in double quotations marks" (Author's Surname, Year of Publication, p. – followed by page number – in brackets)

Archive material/ Special collections

Format:

Author - Surname, Initials (Year of publication - in brackets). Title-in italics or underlined [Material type-in square brackets]. Collection, shelfmark. Archive/Library, City.

Bibliography example:

Turner, P.S. (1906). Image of an Artist [Manuscript]. Holden Collection. 600. Holden Library. London.

In-text example:

(Turner, 1906)

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Atlas

Atlases can be referenced in the same format as a book. If no author is available then the title of the Atlas can be used.

Format:

Atlas with Author:

Author/Originator - Surname, Initials. (Year of publication – in brackets). Title of Atlas in italics or underlined. Series title and volume - if available. Edition - if not the first. Place of Publication: Publisher.

Atlas- no Author:

Title of Atlas – in italics or underlined. (Year of publication - in brackets). Series title and/or volume - if available. Edition - if not the first. Place of publication: Publisher.

Bibliography example:

Wagner, S A. (2004). Color Atlas of the Autopsy. Boca Raton: CRC Press

The Times Concise Atlas of the World. (2000). 8th Ed. London: Times Books

In-text example:

(Wagner, 2004)

As can be seen in The Times Concise Atlas of the World (2000)

Author: Contribution/ Chapter in a Book

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web the rate determining slower step gives a rate law showing second order dependence on the $[A]^2$ concentration and the sum of the two elementary equations gives the overall net reaction in general when the rate determining slower step is the first step in the reaction mechanism the rate law for the overall reaction is the same as the rate

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web rate laws and reaction order the relation between the rate of a reaction and the concentrations of reactants is expressed by its rate law for example the rate of the gas phase decomposition of dinitrogen pentoxide $2\text{N}_2\text{O}_5 \rightarrow 4\text{NO}_2 + \text{O}_2$ has been found to be directly proportional to the concentration of N_2O_5 text rate k N

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web the rate determining step limits the overall rate and therefore determines the rate law for the overall reaction introduction multistep reactions one of the most important applications of kinetics is to the study of reaction mechanisms or the sequences of steps by which chemical reactions occur

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web the initial rate law is then measured for each of the reactions consider the reaction between nitrogen monoxide gas and hydrogen gas to form nitrogen gas and water vapor the following data were collected for this reaction at 1280 text o text c see table below notice that the starting concentrations of ce no and ce h

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web determination of a rate law and activation energy in this experiment glass beakers pipettes and a hotplate were utilized to set up the procedures a solution of deionized water a buffer potassium iodide KI sodium thiosulfate $\text{Na}_2\text{S}_2\text{O}_3$ starch and hydrogen peroxide H_2O_2 were used in order to calculate the rate of reaction

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web oct 27 2022 the rate law $\text{rate} = k[\text{H}_2\text{O}][\text{OH}^-]$ describes a reaction that is first order in H_2O first order in OH^- and second order overall example 12 3 1 writing rate laws from reaction orders an experiment shows that the reaction of

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web rate laws sometimes called differential rate laws or rate equations are mathematical expressions that describe the relationship between the rate of a chemical reaction and the concentration of its reactants as an example consider the reaction described by the chemical equation $\text{a} + \text{b} \rightarrow \text{products}$

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web determining the rate law page 1 page 2 page 3 kinetics experiments the goal of a kinetics experiment is to measure the concentration of a species at a particular time during a reaction so that a rate law can be determined

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web core concepts some reactions will go fast and some will go slow the speed of the reaction is its reaction rate which is

dictated by a rate law in this article we will learn about reaction rates rate laws the rate constant and the reaction order

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web 12 1 in any proceedings tried in any court of record for the recovery of any debt or damages the court may if it thinks fit order that there shall be included in the sum for which judgment is given interest at such rate as it thinks fit on the whole or any part of the debt or damages for the whole or any part of the period between the

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