

Answer Key

Unit 1 Biochemistry

Answers to Unit 1 Preparation Questions Assessing Student Readiness (Student textbook pages 4–7)

1. Sketches should depict a central atomic nucleus surrounded by four valence electrons (or six electrons in total), or six neutrons and six protons within the nucleus.
2. b
3. a
4. metal, non-metal ion
5. **a.** Since the atoms in a glucose molecule share valence electrons, the bonds between these atoms must be covalent.
b. Glucose is a stable molecule because its covalent bonds are stable and strong.
6. **a.** water—molecular, atoms (both non-metals) are joined by covalent bonds
b. carbon dioxide—molecular, atoms (both non-metals) are joined by covalent bonds
c. glucose—molecular, atoms (two non-metals and uncharged hydrogen) are joined by covalent bonds
d. sodium chloride—ionic, ions (a metal and a non-metal) joined by an ionic bond
e. calcium phosphate—ionic, ions (a metal and a polyatomic ion) joined by ionic bonds
f. methane—molecular, atoms (a non-metal and uncharged hydrogen) joined by covalent bonds
g. oxygen—molecule or molecular element, atoms joined by double covalent bond, but not a molecular compound because it is formed from only one element (two oxygen atoms)
h. ammonia—molecular, atoms (a non-metal and uncharged hydrogen) joined by covalent bonds
7. **a.** chloride
b. sulphate/sulfate
c. magnesium ion
d. copper (II) ion
8. **a.** OH^-
b. S^{2-}
c. Al^{3+}
d. Fe^{3+}
9. H_2O or water
10. $\text{water} \rightarrow \text{hydrogen} + \text{oxygen}$
11. d
12. sodium hydroxide + hydrochloric acid $\rightarrow$
water + sodium chloride
13. **a.** Since the pH of the solution increased, the student must have added a base to the solution.
b. One of: When the second substance was added, it released hydroxide ions (OH^-), which combined with protons (H^+) to produce water. The second substance accepted protons (H^+) from the solution.
14. **a.** decomposition
b. neutralization
c. decomposition
d. synthesis
e. neutralization
f. synthesis
g. decomposition
h. synthesis
i. synthesis
j. synthesis
15. pH 7
16. c
17. One of: enzymes, cell structure, membrane transport, muscle
18. A molecule is a substance composed of two or more atoms that are bonded together covalently. A macromolecule is a large, complex molecule. Usually, the subunits of a macromolecule are small molecules bonded together.
19. An enzyme is a macromolecule that speeds up chemical reactions.
20. c
21. c
22. b

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S Nieto



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