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b) propyne c) ethyne a) Butyne d) pentyne

2) The loss of \_\_\_\_\_ and \_\_\_\_ atoms from the double-bonded carbon atoms leads to alkyne formation.

 a) Oxygen, halogen
 b) oxygen, hydrogen
 c) hydrogen, d) carbon, halogen halogen

# What is the reagent used and appropriate temperature required to prepare ethene from ethyl alcohol?

a) Diluted H<sub>2</sub>SO<sub>4</sub>, 160°C

b) Concentrated H<sub>2</sub>SO<sub>4</sub>, 170°C

c) Diluted HCl, 170°C

d) Concentrated HCl, 160°C

# 4) \_\_\_\_\_ on heating with alcoholic potash loses one molecule of hydrogen halide to give alkene.

a) Halogen

b) Haloalkanes

c) Haloalkynes

d) None of the above

#### 5) The IUPAC name of the compound H2C=CH-CH(CH3)2:

a) 2-Vinylpropene

b) 1,1-dimethyl-2-peopene

c) 3-Methyl-1-butene

d) 1-Isopropyl ethylene

#### 6) Ethylbenzene reacts with Cl<sub>2</sub> in sunlight to give:

a) I-Chloro-1-phenylethane

b) 1-Chloro-2-phenylethane

c) 4-Chloroethylbenzene

d) 3-Chloroethylbenzene

#### 7) Benzene Can be :

a) Discharge the color of KMNO<sub>4</sub>

b) Discharge the color of bromine water

c) Polymerize

d) Undergo substitution reactions

#### 8) Which of the following reactions will yield 2, 2 -dibromopropane?

a) HC == CH + 2HBr -->

b) H<sub>3</sub>C — ⊆ == CH<sub>2</sub> + HB<sub>F</sub> — ►

c) H<sub>3</sub>C — C == CH + 2HBr — ► d) H<sub>3</sub>C — C == CHBr + HBr — ►

Give the structure of reactants :

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