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THE OFFICE OF THE ACADEMIC REGISTRAR ANNOUNCEMENT

PRE-ENTRY EXAMINATION FOR ADMISSION TO THE BACHELOR OF LAWS FOR 2023/2024 ACADEMIC YEAR

The Academic Registrar Makerere University announces that the Pre-Entry Examination for admission to the Bachelor of Laws for 2023/2024 will be held on **Saturday 29th April, 2023.**

To be eligible to sit the examination, Applicants must possess the minimum admission requirements of the University, and any one of the following specific requirements:

- i) **'A' Level leavers:** Must hold an 'O' Level Certificate (UCE) or its equivalent with at least 5 Passes plus 'A' Level Certificate (UACE) or its equivalent having obtained a minimum of 13 points (Males) and 12 points (for Females) in any 'A' Level combination for both Humanities and Sciences. Candidates who did 'A' Level before 2013 should have 15 and 14 points for male and female applicants respectively.
- ii) **Diploma Holders:** Must possess at least a 2nd Class or Credit Diploma from a recognised institution. Certified Copies of Diploma Transcripts (No Provisional Results) from the Awarding Institutions must be attached to the Application Forms.
- iii) **Mature Age Applicants:** Must have passed the Makerere University Mature Age Entry Examination (for Bachelor of Laws) for either October, 2021 or December, 2022.
- iv) **Degree Holders:** Must be graduates in any Discipline from a Chartered University. Certified Copies of Degree Transcripts (No Provisional Results) from the Awarding Institutions must be attached to the Application Forms. (Degree holders are not eligible for government sponsorship)

N.B.

- i) **ALL APPLICANTS WISHING TO BE ADMITTED TO THE BACHELOR OF LAWS PROGRAMME OFFERED BY MAKERERE UNIVERSITY ARE REQUIRED TO SIT AND PASS THE PRE-ENTRY EXAMINATION OF MAKERERE UNIVERSITY.**
- ii) **ALL APPLICANTS INTERESTED IN PRIVATE SPONSORSHIP FOR DAY (LAW), AND AFTERNOON (LLA) WILL APPLY FOR ADMISSION WHEN THE ADVERT COMES OUT.**
- iii) **ALL APPLICANTS WHO HOLD DIPLOMAS AND DEGREES MUST PRESENT A CERTIFIED COPY OF THE ACADEMIC TRANSCRIPT. PROVISIONAL RESULTS SHALL NOT BE ACCEPTED TO REGISTER FOR THIS EXAMINATION.**
- iv) **APPLICANTS SHOULD NOTE THAT ADMISSION IS BASED ON PERFORMANCE IN THE PRE-ENTRY EXAMINATION AND INTAKE CAPACITIES FOR PRIVATE**

Questions For The 2015 2016 Law Pre Entry Exam

R Barnett



Questions For The 2015 2016 Law Pre Entry Exam:

Unveiling the Magic of Words: A Overview of "**Questions For The 2015 2016 Law Pre Entry Exam**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is actually awe-inspiring. Enter the realm of "**Questions For The 2015 2016 Law Pre Entry Exam**," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book is central themes, examine its distinctive writing style, and assess its profound affect the souls of its readers.

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web k sp of calcium hydroxide 1 author j m mccormick last update october 13 2013 introduction an equilibrium constant K is related to ΔG for a process through eqn 1 and as long as ΔH and ΔS are independent of temperature eqn 1 can be combined with eqn 2 to determine ΔH and ΔS from the temperature dependence of K

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web in today s experiment you will determine the solubility product ksp of calcium hydroxide ca oh 2 by measuring the concentration of ca oh 2 in a saturated solution calcium hydroxide is a sparingly soluble salt that dissolves according to the following reaction the solubility product expression for this reaction is

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web expert verified chem 182 determining the ksp of calcium hydroxide calcium hydroxide is an ionic solid that is sparingly soluble in water a saturated aqueous solution of ca oh is represented in equation form as shown below ca oh s $\rightleftharpoons$ ca aq 2oh aq the solubility product expression describes in mathematical terms the equilibrium

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web use the concentration of calcium hydroxide to calculate the concentration of calcium ions and hydroxide ions in the solution ca oh 2 $\rightarrow$ ca 2 2oh 5 x 10 2 m calcium ions 5 x 10 2 m 2 0 m hydroxide ions calculate calcium hydroxide ksp using the concentrations of calcium ions and hydroxide ions

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web the chemical equation used was $\text{Ca(OH)}_2 \text{ s} + 2\text{H}^+ \text{ aq} \rightarrow \text{Ca}^{2+} \text{ aq} + 2\text{H}_2\text{O}$ the solubility product was determined to be $K_{\text{sp}} \text{ Ca(OH)}_2$ from the titration which was made by dripping a concentrated solution of calcium hydroxide into hydrochloric acid the calcium hydroxide was filtered before starting the titration

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we will filter the solid calcium hydroxide from the solution and then titrate the dissolved calcium hydroxide in solution with a known concentration of hydrochloric acid using titrimetric calculations we will determine the Ca^{2+} and OH^- concentrations in solution

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we calculate the ksp for calcium hydroxide $\text{ksp Ca(OH)}_2 = 5 \times 10^{-6}$ find the accepted value of the ksp for calcium hydroxide and compare it with your value discuss the discrepancy and suggest possible sources of experimental error
accepted value for ksp for sodium hydroxide is 5×10^{-6}