

REPORT FOR EXPERIMENT 15

Quantitative Preparation of Potassium Chloride

- A. Write the balanced equation for the reaction between K_2CO_3 and HCl :
- B. **Experimental Data and Calculations:** Record all measurement to the precision of the balance and remember to use the proper number of significant figures in all calculations. (The number 0.004 has only *one* significant figure.)
- 1. Mass of empty evaporating dish _____
 - 2. Mass of dish and dry K_2CO_3 _____
 - 3. Mass of dish and residue (KCl) after first heating _____
 - 4. Mass of dish and residue (KCl) after second heating _____
 - 5. Mass of dish and residue (KCl) after third heating (if necessary) _____
 - 6. Mass of potassium carbonate _____

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Scientific and Technical Aerospace Reports ,1969 Nuclear Science Abstracts ,1976 **The Journal of Industrial and Engineering Chemistry** ,1912 **Subject Index to Unclassified ASTIA Documents** Defense Documentation Center (U.S.),1960 **ERDA Energy Research Abstracts** United States. Energy Research and Development Administration,1977-03 *Government Reports Announcements* ,1974 **Government Reports Announcements & Index** ,1984-05

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