

Qualitative Analysis and Chemical Bonding

Concepts:

- Chemical bonding
- Covalent bonding
- Ionic bonding
- Metallic bonding
- Crystal & chemical properties
- Electronegativity

Introduction

- Study of atoms, molecules, crystal structures, properties
- In order to understand these, all atoms/molecules are made of electrons, protons, neutrons, etc. So, the study of atoms, molecules, crystal structures, properties, etc. is the study of the structure of matter, which is the study of the structure of the universe.

Atomic Structure

- Atoms are made of three particles: electrons, protons, neutrons. Electrons are negatively charged, protons are positively charged, and neutrons are neutral.
- Electrons are present in all atoms, but protons and neutrons are present only in some atoms.
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Chemical Bonding

Chemical bonding is the process by which atoms combine to form molecules or crystals. It is the study of the forces that hold atoms together in a molecule or crystal.



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