

Qualitative Analysis and Chemical Bonding

Concepts:

- Chemical bonding
- Covalent bonding
- Ionic bonding
- Metallic bonding
- Intermolecular bonding
- Physical & chemical properties
- Electronegativity

Introduction

- Study of atoms, molecules, and their interactions: **Chemistry**
- In order to understand these, all atoms/molecules are made of electrons, protons, and neutrons. The study of these particles is called **Physics**.
- In order to study a system, we need to know its structure, energy, and how it interacts with its environment. This is called **Thermodynamics**.

- **Chemical bonding** is the process by which atoms combine to form molecules. It is the study of the forces that hold atoms together in a molecule.
- There are three types of chemical bonding: **ionic bonding**, **covalent bonding**, and **metallic bonding**.
- **Ionic bonding** is the result of the electrostatic attraction between oppositely charged ions.
- **Covalent bonding** is the result of the sharing of electron pairs between atoms.
- **Metallic bonding** is the result of the delocalized electrons in a metal lattice.



Green sphere: metal atom
Purple sphere: metal atom

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Qualitative Analysis And Chemical Bonding Lab Teacher

Raffaela Di Napoli



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