



Qualitative Analysis and Chemical Bonding

Concepts:

Chemical bonds, structure-property relationships, ionic bonds, covalent bonds, polar covalent bonds, nonpolar covalent bonds, metallic bonds

Use the following recommendations to increase and/or decrease the challenge difficulty for your students.

Short-on-time Inquiry Lab:

Students follow a detailed procedure to test for six properties of six compounds to determine the type of bonding or interparticle force that holds the constituent particles in each compound together. Students answer questions meant to probe their understanding of the structure-property relationship. (30 min.)

Guided Inquiry Lab:

Students follow a detailed procedure to test for six properties of twelve compounds to determine the type of bonding or interparticle force that holds the constituent particles in each compound together. Students answer questions meant to probe their understanding of the structure-property relationship. (50 min.)

Open Inquiry Lab:

Following an introduction to the fundamental types of bonding, students must develop a procedure to characterize an unknown solid as ionic, polar covalent, nonpolar covalent, or metallic. Students then answer questions meant to probe their understanding of the structure-property relationship. (50–90 min.)

Advanced Inquiry Lab:

In the absence of an introductory activity students must develop a procedure to characterize an unknown solid as ionic, polar covalent, nonpolar covalent, or metallic. (60–90 min.)

Outcomes:

Students come to understand that a material's properties are linked to the material's structure and bonding.

Associated Phenomena:

Why are some materials good for certain applications but not others?

Standards

Science & Engineering Practices	Disciplinary Core Ideas	Crosscutting Concepts
Planning and carrying out investigations Analyzing and interpreting data Constructing explanations and designing solutions	HS-PS1.A: Structure and Properties of Matter	Patterns Structure and function Cause and effect

Performance Expectations

HS-PS1-3: Plan and conduct an investigation to gather evidence to compare the structure of substances at the bulk scale to infer the strength of electrical forces between particles.

Qualitative Analysis And Chemical Bonding Flinn

Tao Wei



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academic libraries in the UK and Ireland **Chemical Principles with Qualitative Analysis** William L. Masterton, Emil J. Slowinski, Conrad L. Stanitski, 1986 **Government Reports Announcements & Index** ,1984 **Complementary**

Bonding Analysis Simon Grabowsky, 2021-04-06 As chemical bonds are not observable there are various theories and models for their description This book presents a selection of conceptually very different and historically competing views on chemical bonding analysis from quantum chemistry and quantum crystallography It not only explains the principles and theories behind the methods but also provides practical examples of how to derive bonding descriptors with modern software and of how to interpret them **Science Citation Index** ,1975 Vols for 1964 have guides and journal lists

Brennstoff-Chemie ,1963 **Chemical Bonding Analysis of Solids in Position Space** Alexey I. Baranov, 2015

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Qualitative Analysis Durward C. Layde, Daryle Hadley Busch, 1968 **Qualitative Analysis and Chemical Equilibrium** ,1937 **Qualitative Analysis and Chemical Equilibrium** Thorfin Rusten Hogness, Warren Charles Johnson, 1937

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