



**FEDERAL PUBLIC SERVICE COMMISSION
SPECIAL COMPETITIVE EXAMINATION-2023 FOR
RECRUITMENT TO POSTS IN RS-17 UNDER THE FEDERAL
GOVERNMENT**

Roll Number



CHEMISTRY, PAPER-I

TIME ALLOWED: THREE HOURS	(PART-I MCQs)	MAXIMUM MARKS: 20
PART-I (MCQs) : MAXIMUM 30 MINUTES	(PART-II)	MAXIMUM MARKS: 80
NOTE: (i) First attempt PART-I (MCQs) on separate OMR Answer Sheet which shall be taken back after 30 minutes.		
(ii) Overwriting/cutting of the options/answers will not be given credit.		
(iii) There is no negative marking. All MCQs must be attempted.		

PART-I (MCQs) (COMPULSORY)

Q.1. (i) Select the best option/answer and fill in the appropriate Box ☐ on the OMR Answer Sheet.(20x1=20)
(ii) Answers given anywhere else, other than OMR Answer Sheet, will not be considered.

- Write the values for l , m , and n for $\Psi_{3,1,0}$?
(A) 1, 3, 0 (B) 3, 1, 0 (C) 0, 3, 1 (D) None of these
- What is the magnetic quantum number of the orbital $2p_x$?
(A) 1 (B) ± 1 (C) -1 (D) None of these
- The weight of silver displaced by a quantity of electricity which displaces 5600ml of O_2 at STP will be:
(A) 5.4g (B) 10.8g (C) 54.9g (D) None of these
- On electrolysis of dilute sulphuric acid using platinum electrodes, the product obtained at the anode will be:
(A) Hydrogen (B) Oxygen (C) Hydrogen sulphide (D) None of these
- The coefficient of performance of a refrigerator is 5. If the temperature inside of said refrigerator is $-20^\circ C$. Calculate the temperature of the surrounding where it releases heat.
(A) $11^\circ C$. (B) $21^\circ C$. (C) $31^\circ C$. (D) None of these
- When an ideal monatomic gas is heated at constant pressure, the fraction of heat energy that increases the internal energy of the gas is:
(A) $2/5$ (B) $3/5$ (C) $3/7$ (D) None of these
- If the concentration of material X is doubled and that of Y is halved, how many times will the rate of the elementary reaction $3X + Y \rightarrow X_2Y$ change?
(A) $r_2 = 4.5r_1$ (B) $r_2 = 5r_1$ (C) $r_2 = 2r_1$ (D) None of these
- The chemical reaction, $2O_3 \rightarrow 3O_2$ Proceeds as
 $O_3 \rightleftharpoons O_2 + [O]$ (fast)
 $[O] + O_3 \rightarrow 2O_2$ (slow)
The rate law expression will be:
(A) Rate = $k [O] [O_3]$ (B) Rate = $k [O_3]^2 [O_2]^{-1}$ (C) Rate = $k [O_3]^2$ (D) None of these
- Which one of the following is an example of adsorption?
(A) Ammonia in contact with water (B) Anhydrous $CaCl_2$ with water
(C) Silica gel in contact with water vapours (D) None of these
- The movement of colloidal particles towards the oppositely charged electrodes on passing electric current is known as:
(A) Tyndall effect (B) Cataphoresis (C) Brownian movement (D) None of these
- The process of separating a crystallloid, from a colloid by filtration is called:
(A) Emulsification (B) Dialysis (C) Coagulation (D) None of these
- Which of the following is used as a carrier gas in gas chromatography?
(A) Carbon dioxide (B) Oxygen (C) Helium (D) None of these
- What is Eluent?
(A) Liquid solution (B) Liquid solution that is a result from Elution
(C) Solute (D) None of these
- Which of the following HPLC detectors is used as a bulk property or general purpose detector?
(A) Electrochemical detector (B) Fluorescence detector
(C) UV-Visible detector (D) None of these
- Barcin oxime is known as:
(A) Dioxime (B) Capron (C) Cupric acid (D) None of these
- In-crystal field theory (CFT), the energy gap between t_{2g} and e_g sets is denoted by:
(A) $10Dq$ (B) $20Dq$ (C) $0Dq$ (D) None of these
- The chemical formula of caustic potash is:
(A) $NaOH$ (B) $Ca(OH)_2$ (C) NH_4OH (D) None of these
- Which one of the following elements shows the maximum number of different oxidation states in its compounds?
(A) Ba (B) Am (C) Cd (D) None of these

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Mark Steyvers



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Table of Contents Pyc37sample Of Examination Paper

1. Understanding the eBook Pyc37sample Of Examination Paper
 - The Rise of Digital Reading Pyc37sample Of Examination Paper
 - Advantages of eBooks Over Traditional Books
2. Identifying Pyc37sample Of Examination Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Pyc37sample Of Examination Paper
 - User-Friendly Interface
4. Exploring eBook Recommendations from Pyc37sample Of Examination Paper
 - Personalized Recommendations
 - Pyc37sample Of Examination Paper User Reviews and Ratings
 - Pyc37sample Of Examination Paper and Bestseller Lists
5. Accessing Pyc37sample Of Examination Paper Free and Paid eBooks
 - Pyc37sample Of Examination Paper Public Domain eBooks
 - Pyc37sample Of Examination Paper eBook Subscription Services
 - Pyc37sample Of Examination Paper Budget-Friendly Options

6. Navigating Pyc37sample Of Examination Paper eBook Formats
 - ePub, PDF, MOBI, and More
 - Pyc37sample Of Examination Paper Compatibility with Devices
 - Pyc37sample Of Examination Paper Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Pyc37sample Of Examination Paper
 - Highlighting and Note-Taking Pyc37sample Of Examination Paper
 - Interactive Elements Pyc37sample Of Examination Paper
8. Staying Engaged with Pyc37sample Of Examination Paper
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Pyc37sample Of Examination Paper
9. Balancing eBooks and Physical Books Pyc37sample Of Examination Paper
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Pyc37sample Of Examination Paper
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Pyc37sample Of Examination Paper
 - Setting Reading Goals Pyc37sample Of Examination Paper
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Pyc37sample Of Examination Paper
 - Fact-Checking eBook Content of Pyc37sample Of Examination Paper
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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