

Graphing Periodic Trends Lab

Name _____ Hr _____

PURPOSE: To determine how certain properties are periodic when the elements are arranged in periods or groups.

DISCUSSION: There are many periodic trends that occur as you cross a period or go down a group/family on the Periodic Table. Graphing various elemental properties will help to illustrate these trends.

REFERENCE:

Angstrom - A unit of distance equal to one ten-billionth of a meter.

Atomic Radius - The distance from the center of the nucleus to the valence electron energy level.

Ionization energy - The energy required to remove an electron from an atom.

DATA

Properties of the First 18 Elements

Element	Atomic Number	Atomic Radius (Angstroms)	Ionization Energy
H	1	0.79	314
He	2	0.49	567
Li	3	2.05	124
Be	4	1.4	215
B	5	1.17	191
C	6	0.91	260
N	7	0.75	335
O	8	0.65	314
F	9	0.57	402
Ne	10	?	?
Na	11	2.23	119
Mg	12	1.72	176
Al	13	1.82	138
Si	14	1.46	168
P	15	1.23	242
S	16	1.09	239
Cl	17	0.97	299
Ar	18	?	?

Using the data above, make a line graph the following data. Draw a dark vertical line to separate the 3 periods. Give each graph a title and label both the x-axis and the y-axis. Connect your points. Predict the plot points for elements with question marks.

1. Atomic Radius (y) vs. Atomic Number (x) for the first 3 periods (18 elements – 3 graphs)
2. Ionization energy (y) vs. Atomic Number (x) for the first 3 periods (18 elements – 3 graphs)

Periodic Trends Lab Answers

Mary Louise Turgeon



Periodic Trends Lab Answers:

CBSE Science Chapterwise Case Study Class 10 Priti Singhal, 2024-11-17 This book is structured to align with the latest syllabus and curriculum guidelines ensuring that the content is both relevant and rigorous Each chapter begins with a clear set of learning objectives providing a roadmap for students to understand what they will achieve by the end of the chapter We have included numerous diagrams illustrations and real life examples to make complex concepts more accessible and engaging

Hands-On General Science Activities With Real-Life Applications Pam Walker, Elaine Wood, 2008-04-21 In this second edition of Hands On General Science Activities with Real Life Applications Pam Walker and Elaine Wood have completely revised and updated their must have resource for science teachers of grades 5 12 The book offers a dynamic collection of classroom ready lessons projects and lab activities that encourage students to integrate basic science concepts and skills into everyday life

Exploring General Chemistry in the Laboratory Colleen F. Craig, Kim N. Gunnerson, 2017-02-01 This laboratory manual is intended for a two semester general chemistry course The procedures are written with the goal of simplifying a complicated and often challenging subject for students by applying concepts to everyday life This lab manual covers topics such as composition of compounds reactivity stoichiometry limiting reactants gas laws calorimetry periodic trends molecular structure spectroscopy kinetics equilibria thermodynamics electrochemistry intermolecular forces solutions and coordination complexes By the end of this course you should have a solid understanding of the basic concepts of chemistry which will give you confidence as you embark on your career in science

Using Constructivism in Teaching AP Chemistry Avidah Lotfi, 2004 *Developing and Sustaining a Research-supportive Curriculum* Kerry K. Karukstis, Timothy E. Elgren, 2007 This compendium of successful curricular and institutional practices to develop critical research skills emphasized the importance of the collective efforts of the undergraduate community to integrate research and education By collecting and disseminating a variety of mechanisms that are effective means of creating a research supportive undergraduate curriculum the Council on Undergraduate Research aims to encourage faculty and institutions to continue to seek creative useful and significant ways to promote learning through research Publisher s description

Chemistry in the Laboratory James M. Postma, Julian L. Robert, J. Leland Hollenberg, 2004-03-12 This clearly written class tested manual has long given students hands on experience covering all the essential topics in general chemistry Stand alone experiments provide all the background introduction necessary to work with any general chemistry text This revised edition offers new experiments and expanded information on applications to real world situations

Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 **O-level Chemistry Challenging Drill Questions (Concise) (Yellowreef)** Thomas Bond, Chris Hughes, 2013-11-07 questions from top schools since 2003 complete encyclopedia of over 350 questions complete answer keys full set of step by step solution approaches available separately complete edition and concise edition eBooks available

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Allan Beran,2010-11-01 This new edition of the Beran lab manual emphasizes chemical principles as well as techniques The manual helps students understand the timing and situations for the various techniques The Beran lab manual has long been a market leading lab manual for general chemistry Each experiment is presented with concise objectives a comprehensive list of techniques and detailed lab intros and step by step procedures *Parallel Curriculum Units for Science, Grades 6-12* Jann

H. Leppien,Jeanne H. Purcell,2011-02-15 Based on the best selling book The Parallel Curriculum this resource deepens teachers understanding of how to use the Parallel Curriculum Model PCM to provide rigorous learning opportunities for students in science grades 6 12 This collection of sample units and lessons within each unit were developed by experienced teachers and demonstrate what high quality curriculum looks like within a PCM framework Ideal for use with high ability students the units revolve around genetics the convergence of science and society the integration of English and Biology and the Periodic Table Lessons include pre and post assessments **Prentice Hall Science Explorer** Michael J. Padilla,2002

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