Elements of the Sea

fanic equations

 If 2 solutions react to form a solid, a precipitation reaction occurs. Eq. Ag+(ag) + No3-(ag) + No+(ag) + Er-(ag) goes to AgBr(s) + No+9agO + NO3-(ag)

There cancel spectators

Ag+(aq) + Br-(aq) goes to AgBr(s)

- Always include state symbols.
- Only split into ions if agueous and ionic.

tonic solids

- Cations and anions are held together by strong electrostatic attractions in a grant management
 - Each sodium ion is surrounded by the thioride ions to the perdination is in a simple public structure.
 - lonic solids are hard and have a high melting point due to strong electrostatic attractions between ions.
- a If ionic crystals contain water, the Deciminate Colombian water molecules fit into the lattice the same way as regular ions. This is called the water of crystallisation. These crystals are said to be hydrated.
 - If these hydrated crystals are heated, the water molecules are driven off and you are left with an arrhydroun solid.

lionic substances in solution

 As lonic substances dissolve in water, the lons become surrounded by water molecules and they spread throughout the solution. The hydrated ions are randomly arranged and behave independently.

Salters Elements Of Sea Module Test

L Cohen

Salters Elements Of Sea Module Test:

Unveiling the Magic of Words: A Report on "Salters Elements Of Sea Module Test"

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