

CHAPTER 7 REVIEW*Chemical Formulas and Chemical Compounds***SECTION 7-1****SHORT ANSWER** Answer the following questions in the space provided.

1. C In a Stock name such as iron(III) sulfate, the roman numeral tells us _____
(a) how many atoms of Fe are in one formula unit
(b) how many sulfate ions can be attached to the iron atom
(c) the charge on each Fe ion
(d) the total positive charge of the formula unit
2. C The result of changing a subscript in a correctly written chemical formula is to _____
(a) change the number of moles represented by the formula
(b) change the charges on the other ions in the compound
(c) change the formula so that it no longer represents the compound it previously represented
(d) have no effect on the formula
3. The explosive TNT has the molecular formula $C_7H_5(NO_2)_3$.
4 elements a. How many elements make up this compound?
6 oxygen atoms b. How many oxygen atoms are present in one molecule of $C_7H_5(NO_2)_3$?
21 atoms c. How many atoms in total are present in one molecule of $C_7H_5(NO_2)_3$?
 4.2×10^{24} atoms d. How many atoms are present in a sample of 2×10^{23} molecules of $C_7H_5(NO_2)_3$?
4. How many atoms are present in each of these formula units?
11 atoms a. $Ca(HCO_3)_2$
45 atoms b. $C_{12}H_{22}O_{11}$
10 atoms c. $Fe(ClO_2)_3$
9 atoms d. $Fe(ClO_3)_2$
5. N_2O_5 a. What is the formula for the compound dinitrogen pentoxide?
carbon(IV) sulfide b. What is the Stock name for the covalent compound CS_2 ?
 H_2SO_3 c. What is the formula for sulfurous acid?
phosphoric acid d. What is the name for the acid H_3PO_4 ?

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Emilie Sanchez

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Table of Contents Review Solutions Section 13 2 Modern Chemistry

1. Understanding the eBook Review Solutions Section 13 2 Modern Chemistry
 - The Rise of Digital Reading Review Solutions Section 13 2 Modern Chemistry
 - Advantages of eBooks Over Traditional Books
2. Identifying Review Solutions Section 13 2 Modern Chemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Review Solutions Section 13 2 Modern Chemistry
 - User-Friendly Interface
4. Exploring eBook Recommendations from Review Solutions Section 13 2 Modern Chemistry
 - Personalized Recommendations
 - Review Solutions Section 13 2 Modern Chemistry User Reviews and Ratings
 - Review Solutions Section 13 2 Modern Chemistry and Bestseller Lists
5. Accessing Review Solutions Section 13 2 Modern Chemistry Free and Paid eBooks
 - Review Solutions Section 13 2 Modern Chemistry Public Domain eBooks
 - Review Solutions Section 13 2 Modern Chemistry eBook Subscription Services
 - Review Solutions Section 13 2 Modern Chemistry Budget-Friendly Options
6. Navigating Review Solutions Section 13 2 Modern Chemistry eBook Formats

- ePub, PDF, MOBI, and More
- Review Solutions Section 13 2 Modern Chemistry Compatibility with Devices
- Review Solutions Section 13 2 Modern Chemistry Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Review Solutions Section 13 2 Modern Chemistry
 - Highlighting and Note-Taking Review Solutions Section 13 2 Modern Chemistry
 - Interactive Elements Review Solutions Section 13 2 Modern Chemistry
- 8. Staying Engaged with Review Solutions Section 13 2 Modern Chemistry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Review Solutions Section 13 2 Modern Chemistry
- 9. Balancing eBooks and Physical Books Review Solutions Section 13 2 Modern Chemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Review Solutions Section 13 2 Modern Chemistry
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Review Solutions Section 13 2 Modern Chemistry
 - Setting Reading Goals Review Solutions Section 13 2 Modern Chemistry
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Review Solutions Section 13 2 Modern Chemistry
 - Fact-Checking eBook Content of Review Solutions Section 13 2 Modern Chemistry
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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