

JEE-Mains-25-01-2023 (Memory Based) [Shift-2] Chemistry

SECTION - A

Multiple Choice Questions: This section contains 20 multiple choice questions. Each question has 4 choices (1), (2), (3) and (4), out of which ONLY ONE is correct.

Choose the correct answer:

- If [H-] in concentration is increased by a factor of 1000. Then pH?
 - (1) Decreased by 3
 - (2) Increased by 3
 - (3) There is no change in pH
 - (4) Decreased by 1

Answer (1)

- Sol. If [H-] is increased by 1000 times then pH will be decreased by 3.
- Arrange the following elements in increasing order of metallic character
 - Si, K, Mg and Be
 - (1) Si < Mg < Be < K
- (2) Be < Mg < Si < K
- (3) Si < Be < Mg < K
- (4) K < Mg < SI < Be

Answer (3)

- Sol. Based on the electronegativity of the given elements, the correct increasing order of metallic character is Si < Be < Mg < K
- 3. Which of the following has two chiral centres
 - (1) 2- Bromo 3- deutro butane
 - (2) 1- Bromo 2- deutro butane
 - (3) 1- Bromo 3- deutro butane
 - (4) 1- Bromo 4- deutro butane

Answer (1)

Sol.



2 - Bromo - 3 deutro butane has two chiral centres.

- A: Carbon form two oxides CO and CO₂, where CO
 is neutral while CO₂ is acidic.
 - R : CO₂ will combine with water to give carbonic acid and CO is soluble in water
 - [A] and [R] both are correct and [R] is correct explanation of [A]
 - (2) [A] and [R] both are correct and [R] is not correct explanation of [A]
 - (3) [A] is correct white [R] is false
 - (4) [A] is talse while [R] is correct

Answer (2)

- Sol. CO₂ will form carbonic acid with water and it is acidic in nature, while CO is neutral but there is no relation of neutrality with solubility.
- Which of the following element is the weakest reducing agent in aqueous solution.
 - (1) Na

(2) K

(3) LI

(4) Rb

Answer (1)

- Sol. As per the standard reduction potential values, Na is the weakest reducing agent.
- Match List-I with List-II

(List-I) Amine

pK_b (Aqueous medium)

L. Best-BL

(a) Anilline

1. 9.0

(b) Ethanamine

- 2. 3.29
- (c) N-ethylethanamine
- 3. 3.25
- (d) N, N-diethylethanamine 4, 3.0
- 4 30
- (1) $a \rightarrow 1, b \rightarrow 2, c \rightarrow 4, d \rightarrow 3$
- (2) a → 1, b → 4, c → 3, d → 2
- (3) a → 1, b → 2, c → 3, d → 4
- (4) $a \rightarrow 2, b \rightarrow 3, c \rightarrow 4, d \rightarrow 1$

Answer (1)

- Sol. The order of basicity is:
 - c>d>b>a
 - .. pK₀ order is : -+ c < d < b < a

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Christian G. Meyer

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