

Data Table 1 Qualitative Anion Test

Chemical	Anion	Added AgNO ₃	Added HCl
NaBr	Bromide	Thicker consistency, cloudy/ white milky	white/ milky liquid formed
Na ₂ CO ₃	Carbonate	Seperated into milky marbled look	Cloudy air bubbles, acidic- plastic on bottom
NaCl	Chloride	Pale white	Cloudy white chunks found at the bottom
NaI	Iodide	Bright yellow, separated immediately	Color changed to a lighter lime color
Na ₂ SO ₄	Sulfate	No observations detected	Changed to a pearl color
Na ₂ S	Sulfide	Dark brown immediately	Merkier than before and gav off a foul smell
Unknown #103	Carbonate	Milky precipitate	Created air bubbles
Unknown #104	Sulfide	Brown immediately	Gave off a foul smell

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Hervey Hubbard Barber



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Qualitative Anion-cation Analysis Emil J. Margolis, 1962 **Tables for Qualitative Chemical Analysis** Heinrich Will, 1852 **Qualitative Analysis and Ionic Equilibrium** George H. Schenk, 1995-11-10 A supplement for courses with a qualitative analysis component this lab manual contains explanations of the chemistry of metal ions and anions It includes pre lab exercises experiments and lab reports Qualitative Analysis and the Properties of Ions in Aqueous Solution Emil J. Slowinski, William L. Masterton, 1990 This inexpensive qualitative analysis supplement offers maximum flexibility and can accompany general chemistry texts Works well with any general chemistry text where the instructor wants more qualitative analysis in conjunction with regular class work **Semimicro Anion Analysis** Hervey Hubbard Barber, 1948

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