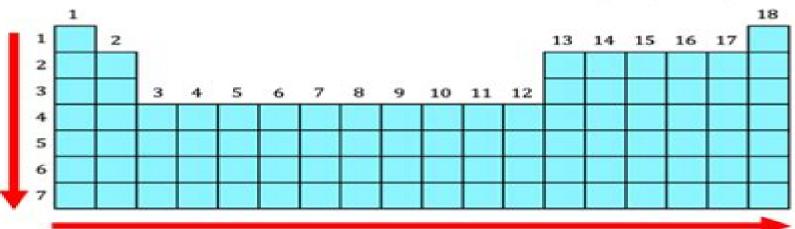
Periodic Trends - Atomic Mass

- The atomic mass describes the mass of each individual element.
- Atomic mass increases moving down each group and right across a period.

<u>Instructions</u>: In the periodic table below, draw two arrows representing the direction that atomic mass <u>increases</u> across groups and periods.



Instructions: In each question, identify the element with the greatest atomic mass:

1) Chlorine (Cl), Iodine (I), Bromine (Br), Fluorine (F)

Iodine

2) Oxygen (O), Carbon (C), Nitrogen (N), Boron (B)

Oxygen

3) Lithium (Li), Silicon (Si), Sulfur (Br), Xenon (Xe)

Xenon

Instructions: In each question, list the elements from least to greatest atomic mass:

4) Magnesium (Mg), Beryllium (Be), Barium (Ba)

Be, Mg, Ba

5) Gallium (Ga), Selenium (Se), Potassium (K)

K, Ga, Se

Periodic Table Trends Answer

American Chemical Society

Periodic Table Trends Answer:

Unveiling the Power of Verbal Beauty: An Mental Sojourn through **Periodic Table Trends Answer**

In a global inundated with monitors and the cacophony of immediate conversation, the profound power and mental resonance of verbal art often diminish in to obscurity, eclipsed by the regular onslaught of noise and distractions. However, located within the musical pages of **Periodic Table Trends Answer**, a captivating function of literary beauty that impulses with organic thoughts, lies an unforgettable journey waiting to be embarked upon. Composed by a virtuoso wordsmith, this mesmerizing opus books visitors on a mental odyssey, softly exposing the latent possible and profound impact embedded within the intricate web of language. Within the heart-wrenching expanse with this evocative analysis, we will embark upon an introspective exploration of the book is central themes, dissect their charming publishing model, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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