

Name: _____
Hour: _____ Date: _____

Chemistry: *The Periodic Table*

Directions: Fill in the blanks on the right with the information in the chart below.

Word List

actinide series	metal
alkali metal	metalloid
alkaline earth metal	Moseley
atomic mass	noble gas
atomic number	nonmetal
family	period
group	periodic law
halogen	periodic table
lanthanide series	transition element

Dmitri Mendeleev developed a chartlike arrangement of the elements called the ____ (1) _____. He stated that if the elements were listed in order of increasing ____ (2) _____, their properties repeated in a regular manner. He called this the ____ (3) _____ of the elements. The arrangement used today, devised by ____ (4) _____, differs from that of Mendeleev in that the elements are arranged in order of increasing ____ (5) _____. Each horizontal row of elements is called a(n) ____ (6) _____. Each vertical column is called a(n) ____ (7) _____, or, because of the resemblance between elements in the same column, a(n) ____ (8) _____.

In rows 4 through 7, there is a wide central section containing elements, each of which is called a(n) ____ (9) _____. Rows 6 and 7 also contain two other sets of elements that are listed below the main chart. These are called the ____ (10) _____ and the ____ (11) _____, respectively. Each of these elements, as well as those in the first two columns at the left end of the chart, is classified as a(n) ____ (12) _____. Each of the elements at the right side of the chart is classified as a(n) ____ (13) _____. Each of the elements between these two main types of elements, having some properties in common with each, is called a(n) ____ (14) _____.

Each of the elements in the column labeled 1 is called a(n) ____ (15) _____. Each of the elements in the column labeled 2 is called a(n) ____ (16) _____. Each of the elements in column 17 is called a(n) ____ (17) _____. Each of the elements in column 18 is called a(n) ____ (18) _____.

Periodic Table Paragraph

**Geoff Martz, Kim Magloire, Theodore
Silver**

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