

Periodic Trends Worksheet

Name: _____ Grade: _____ ID: _____

Directions: Use your notes to answer the following questions.

- 1 Rank the following elements by increasing atomic radius: carbon, aluminum, oxygen, potassium.

Ans: O, C, Al, K

- 2 Rank the following elements by increasing electronegativity: sulfur, oxygen, neon, aluminum.

Ans: Ne, Al, S, O

- 3 Why does fluorine have a higher ionization energy than iodine?

Ans: Fluorine has a smaller atomic radius, so the proton can exert a greater pull on electron (Valence shell is close)

- 4 Why do elements in the same family generally have similar properties?

Ans: They have the same valence shell electron arrangement.

- 5 Indicate whether the following properties increase or decrease from left to right across the periodic table.

- a. atomic radius (excluding noble gases) **Decrease**
- b. first ionization energy **Increase**
- c. electronegativity **Increase**

- 6 What trend in atomic radius occurs across the periodic table? What causes this trend?

Ans: Atomic radius increases down a group because energy levels (shells) are added.

- 7 What trend in ionization energy occurs across a period on the periodic table? What causes this trend?

Ans: Ionization energy increases across a period because as elements become less metallic, it requires more energy to remove an electron.

- 8 Circle the atom in each pair that has the largest radius.

a. Al or B

d. O or F

b. Na or Al

e. Br or Cl

c. S or O

f. Mg or Ca

Periodic Table Trends Answer Key

Yi-Tong Ma

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Periodic Table Trends Answer Key:

Enjoying the Track of Appearance: An Emotional Symphony within **Periodic Table Trends Answer Key**

In a world consumed by monitors and the ceaseless chatter of instant communication, the melodic splendor and emotional symphony produced by the written word usually fade in to the back ground, eclipsed by the persistent noise and interruptions that permeate our lives. Nevertheless, set within the pages of **Periodic Table Trends Answer Key** a charming fictional value filled with organic feelings, lies an immersive symphony waiting to be embraced. Constructed by a masterful musician of language, this fascinating masterpiece conducts visitors on a mental journey, well unraveling the concealed songs and profound affect resonating within each cautiously constructed phrase. Within the depths of this poignant examination, we shall discover the book is central harmonies, analyze its enthralling writing design, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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