Name:

#### Student Exploration: Star Spectra

Vocabulary: absorption spectrum, binary star, blueshift, Cepheid variable, emission spectrum, glant star, nebula, redshift, spectrum, star

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

- What happens when light goes through a prism?

  This band of colors is called a spectrum.
- 2. A rainbow is an example of a spectrum. What is the sequence of colors in a rainbow?

#### Gizmo Warm-up

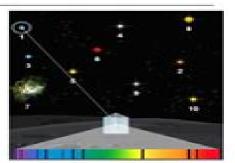
The interior of a **star** produces a continuous spectrum of light, like a rainbow. Cooler gases in the outer layers of the star absorb certain wavelengths of light, causing dark lines to appear in the spectrum. The resulting **absorption spectrum** can tell astronomers a great deal about the star.

 On the Star Spectra Gizmo™, turn on Show labels. Select star 1 to see its absorption spectrum.

How many lines do you see in the spectrum?

2. Drag the Hydrogen spectrum next to the Star spectrum so that the edges line up. Do

some of the lines on the two spectra match up?



- Drag the Helium spectrum next to the Star spectrum. Do some lines match?
- 4. Try out the other available spectra. Do any others have lines that match?
- 5. Which elements have contributed to the spectrum of star 1?

Note: The elements in a star's spectrum do not necessarily reflect what the star is made of. The absorption spectrum is mostly determined by the surface temperature of the star.

Activity A: Classifying stars	Get the Gizmo ready: Check that Show labels is on and Neutral spectra is selected.	

#### Question: How are stars classified?

**Introduction**: Late in the 19th century, Harvard astronomer Edward Pickering wanted to sort and catalog the thousands of star spectra that had been collected by the Harvard Observatory. He hired several women to do the work, paying them 25 cents a day. The most prominent of these women was Annie Jump Cannon, who devised a classification system still used today.

Desiry St.				
Harvard Classification Scheme				
Class	Color	Prominent spectral lines	Surface Temp. (K	
0	Blue	Ionized helium, hydrogen	> 25,000	
В	Blue-white	Neutral helium, hydrogen	11,000 - 25,000	
A	White	Hydrogen, ionized sodium, ionized calcium	7,500 - 11,000	
F	White	Hydrogen, ionized sodium and calcium; neutral sodium and calcium	6,000 - 7,500	
G	Yellow	Neutral sodium and calcium, ionized calcium, ionized iron, ionized magnesium	5,000 - 6,000	
K	Orange	Neutral calcium, neutral iron, neutral magnesium	3,500 - 5,000	
M	Red	Neutral iron, neutral magnesium, and neutral titanium oxide (not shown)	< 3,500	

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Zhe Xu

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#### **Table of Contents Science World Answers April 2014**

- 1. Understanding the eBook Science World Answers April 2014
  - The Rise of Digital Reading Science World Answers April 2014
  - $\circ\,$  Advantages of eBooks Over Traditional Books
- 2. Identifying Science World Answers April 2014
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Science World Answers April 2014
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Science World Answers April 2014
  - Personalized Recommendations
  - Science World Answers April 2014 User Reviews and Ratings
  - Science World Answers April 2014 and Bestseller Lists
- 5. Accessing Science World Answers April 2014 Free and Paid eBooks
  - Science World Answers April 2014 Public Domain eBooks
  - Science World Answers April 2014 eBook Subscription Services

- Science World Answers April 2014 Budget-Friendly Options
- 6. Navigating Science World Answers April 2014 eBook Formats
  - o ePub, PDF, MOBI, and More
  - Science World Answers April 2014 Compatibility with Devices
  - Science World Answers April 2014 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Science World Answers April 2014
  - Highlighting and Note-Taking Science World Answers April 2014
  - Interactive Elements Science World Answers April 2014
- 8. Staying Engaged with Science World Answers April 2014
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Science World Answers April 2014
- 9. Balancing eBooks and Physical Books Science World Answers April 2014
  - Benefits of a Digital Library
  - o Creating a Diverse Reading Collection Science World Answers April 2014
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Science World Answers April 2014
  - Setting Reading Goals Science World Answers April 2014
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Science World Answers April 2014
  - Fact-Checking eBook Content of Science World Answers April 2014
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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