

Sample of Principal Letter of Recommendation:

Date

National Association of Student Councils
c/o NASC Student Leader Certification
1904 Association Drive
Reston VA 20191

To Whom It May Concern:

I am writing today to recommend (***STUDENT NAME***) be named a certified student leader through the NASC Student Leaders Program. In addition to his/her meeting the required criteria, I would like to highlight additional reasons why this student should receive the award.

STUDENT is an outstanding young leader who has made a difference here at ***SCHOOL*** and who is committed to the student council. I have had the pleasure of observing and working with ***STUDENT*** for two years and have witnessed his/her leadership in action.

One of the best examples I can cite that will illustrate what typifies this young man/woman occurred last year when s/he was working with the student council to assist with a local blood drive. The activity became a passion and his/her efforts and ingenuity were directly responsible for signing up over 300 student volunteers to either give blood or recruit their parents and others. It was one of the most successful drives of the year.

As ambassadors for our school our student council members are known to our parents and community. ***STUDENT*** works diligently each year to reflect to the community the pride and spirit we have here at ***NAME*** High School.

It is for these and other similar and compelling reasons that I am recommending ***STUDENT*** for the honor of being recognized by NASC as a Distinguished Student Leader.

Sincerely,

Mr./Ms. Principal

Sample Letter Of Recommendation For High School Student Council

IM Harris



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 $\text{H}_2\text{O}(\text{g}) \rightarrow \text{H}_2(\text{g}) + \frac{1}{2} \text{O}_2(\text{g}) \quad \Delta h = 71.2 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g}) + \frac{7}{2} \text{O}_2(\text{g}) \rightarrow \text{CO}_2(\text{g}) + 3 \text{H}_2\text{O}(\text{g}) \quad \Delta h = 283 \text{ kJ}$

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 $\text{C}_2\text{H}_4(\text{g}) + 3 \text{O}_2(\text{g}) \rightarrow 2 \text{CO}_2(\text{g}) + 2 \text{H}_2\text{O}(\text{l}) \quad \Delta h = 1411 \text{ kJ/mole}$
 $\text{C}_2\text{H}_6(\text{g}) + 7/2 \text{O}_2(\text{g}) \rightarrow 2 \text{CO}_2(\text{g}) + 3 \text{H}_2\text{O}(\text{l}) \quad \Delta h = 1560 \text{ kJ/mole}$
 $\text{H}_2(\text{g}) + 1/2 \text{O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{l}) \quad \Delta h = 285 \text{ kJ/mole}$

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