

PRE-AP CHEMISTRY: Semester 2 Final Review

Final: June 8 (period 7); June 9 (period 6)

Review days after school: Monday June 1, Wednesday June 3, Thursday June 4

*You have come a long way this year! You should celebrate your successes and continue to work hard until the end. Let me know how I can help you!

**You can also use a periodic table and your reference sheets from the semester.

***Your binder contains your work from the whole semester (maybe even the whole year!) You can study from your semester's old worksheets, practice problems within the chapters, labs, quizzes, or tests, or look in my folders for extra worksheets/practice.

I. SEMESTER 1:

- a. Significant figures: rules for measuring, rounding, etc.
 - i. Why do we need them?
 - ii. How can you tell how many sig figs are in a measurement?
- b. Scientific notation, converting metric units
 - i. $5.64 \times 10^3 =$ _____
 - ii. $0.000374 =$ _____
 - iii. $5000 \text{ grams} =$ _____ kg
 - iv. $0.345 \text{ L} =$ _____ mL
- c. Physical/chemical changes (in Ch. 9 again)
- d. Structure of an atom: numbers of protons, neutrons, electrons
 - i. What are isotopes?
 - ii. How do you identify the number of protons, neutrons, and electrons in an atom?
- e. Electron configurations
 - i. Use the Aufbau principle to write e^- configs for any element on P-table
 - ii. Know electron configurations for **ions**; what has changed? Why do atoms become ions?
- f. Arrangement of periodic table
 - i. Names of specific groups on table: halogens, alkali metals, noble gases, metals, nonmetals

II. UNIT 5 IONIC BONDING (Chapter 4/12)

- i. How do positive ions form? Negative ions?
- ii. What is the octet rule? How does ionic bonding happen?
- iii. Naming and writing formulas, including those with transition metals
 1. $\text{MgCl}_2 =$ _____
 2. $\text{KF} =$ _____
 3. Iron (III) oxide = _____
- iv. Common polyatomic ions
 1. $\text{Na}_2\text{CO}_3 =$ _____
 2. Aluminum hydroxide = _____

III. UNIT 5: COVALENT BONDING

- a. What is the octet rule? How does covalent bonding happen?
- b. Single, double, triple bonds + valence electrons
- c. Naming and writing formulas (system of prefixes)
- d. VSEPR: predicting shapes of molecules
- e. Polar/nonpolar: what is it, how do you know?
 - i. Draw Lewis structures for the following molecules and identify the shape and polarity:
 1. H_2O
 2. NH_3
 3. CH_4
 4. SF_6
 5. SO_4^{2-}

IV. UNIT 6/7: CHEMICAL REACTIONS

- a. What is a chemical reaction?

Pre Ap English 8 Semester 1 Exam Review

CollegeBoard

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Pre Ap English 8 Semester 1 Exam Review: Bestsellers in 2023 The year 2023 has witnessed a remarkable surge in literary brilliance, with numerous engrossing novels captivating the hearts of readers worldwide. Lets delve into the realm of bestselling books, exploring the fascinating narratives that have charmed audiences this year. The Must-Read : Colleen Hoover's "It Ends with Us" This poignant tale of love, loss, and resilience has captivated readers with its raw and emotional exploration of domestic abuse. Hoover expertly weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can triumph. Pre Ap English 8 Semester 1 Exam Review : Taylor Jenkins Reids "The Seven Husbands of Evelyn Hugo" This captivating historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reids captivating storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Discover the Magic : Delia Owens "Where the Crawdads Sing" This evocative coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens weaves a tale of resilience, survival, and the transformative power of nature, captivating readers with its evocative prose and mesmerizing setting. These top-selling novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of captivating stories waiting to be discovered. The novel begins with Richard Papan, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a exceptional and thrilling novel that will keep you wondering until the very end. The novel is a cautionary tale about the dangers of obsession and the power of evil.

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