CHAPTER: 17
CHESCHETCONTON OF COUNTS
CLASSIFICATION OF SOLIDS
(i) Crystalline Solids:
•
· Solids in which atoms are arranged
in pattern; structure have arrangement. • Arrangment of atoms repeats 3-
dimensionally throughout the structure.
0 0
example: Metals: like copper conjc compound: NACL
Ionic compound: NACL C
Ceramics : Zirconia -
NOTE: Atoms oxygogenent in
NOTE: Atoms arrangement in crystalline solids can be studied by x-rays diffraction using Bragg's
x-rays diffraction using Bragg's
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Cohesive force between the atoms
IS V. V. X/XYCIP
Hardest Melting
Point → · sharp • v·v·high · definite
v.v. high definite

Physics Chapter17 Assessment Answers

Princeton Review, Steven A. Leduc

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