

Physical Science – 2nd Semester – Final Exam Study Guide – (2012–2013)

1. Physical Science Basics

- What tool(s) would you want to use to find the:
 - Mass of an object? triple beam balance Basic SI Unit? kg
 - Volume of an object? ruler or water displacement method Basic SI Unit? cm³ or ml
 - Length of an object? Ruler Basic SI Unit? meter
 - Density of an object? graduated cylinder Basic SI Unit? ml
- What is the formula for calculating density? mass/volume
 - What is the density of a cube that is 25 g and has a volume of 3 cm³? 8g/cm³
 - Compare the density of solid H₂O to liquid H₂O, and explain what observable effect this has.
solid H₂O less dense than liquid H₂O

2. Chemistry Basics

- Element (def.) substance made up of one kind of atom
- Atom (def.) basic building block of matter
- Compound (def.) a molecule that contains atoms or molecules of two or more different elements
- Molecule (def.) a combination of two or more of the same or different atoms that are bonded together

3. Model of the Atomic Evolution of the Atomic Theory

Atoms are so small that many scientists throughout history have created models to describe them.

Why have atomic theories changed over time? Better technology came around.

Scientist	Year	Major Discovery/ Contribution	Name of atomic model
John Dalton	1802	Each element is made up of small, indivisible particles called atoms. Different elements have atoms of different masses.	Billiard Ball Model
J.J. Thomson	1897	Negatively charged electron	Plum Pudding Model
Niels Bohr	1913	Electrons move <u>fixed orbits</u> around the atom's nucleus in specific layers, or <u>electron shells</u> . Atoms absorb or give off energy when the electrons move from one shell (<u>energy level</u>) to another.	Bohr Model
Ernest Rutherford	1909	1909 /Based on idea that positive reflects positive/ atom mostly empty space; electrons orbit randomly around positively charged nucleus	Planetary Model

4. Periodic Table

- The current periodic table is arranged by increasing atomic number.
- Reading the Periodic Table
 - An element's properties can be predicted from its location in the periodic table.
 - The atomic number represents the number of protons.
 - The atomic mass represents the number of protons and neutrons.

Physical Science Study Guide Semester Exam

Tobias Bleicker



Physical Science Study Guide Semester Exam:

Physics Handbook & Study Guide Grade 11-12 IEB Kevin Smith,2024-02-01 A comprehensive summary of Grade 11 12 Physics Simple logical summaries with example exam questions and work through solutions The book covers the fundamentals of Grade 11 12 Physics and complements the material in any class text **DSST Principles of Physical Science - the Easy Way to Pass!** David C. Wilson,2007-09-06 Yes you can Ace the DSST Principles of Physical Science 1 exam and earn three college semester hours credit in Physical Science without ever having taken the class Simply by turning up on the day and passing the DSST examination SE SG 512 after using this study guide This Guide adequately and concisely covers all of the course work and should lead to a good passing score in the DSST examination Includes 122 sample test questions **X-kit Exam 2004 Physical Science** ,2005 **CHEMISTRY HANDBOOK & STUDY GUIDE Gr11-12 NE** Kevin Smith,2024-02-01 A comprehensive summary of Grade 11 12 Physics Simple logical summaries with example exam questions and work through solutions The book covers the fundamentals of Grade 11 12 Physics and complements the material in any class text *CLEP Official Study Guide* The College Board,2005-08-20 The CLEP Official Study Guide 17th Edition is the only complete source of information about the CLEP exams You get Descriptions of the exams to help students decide which exams to take Sample questions and answers for all 35 exams Valuable test taking tips and information on obtaining credit through CLEP With more than 6 million CLEP exams taken since 1967 the College Level Examination Program is a credit by examination program that has proven to be effective for students advisors and faculty Students can demonstrate their comprehensive subject knowledge gain credit and move on to more advanced topics Advisors can help students plan their undergraduate years by helping them shorten the time needed to complete their degree Faculty can use their undergraduate introductory courses more effectively There are exams in 35 introductory college level subjects and are administered on computers at 1 300 college testing centers Approximately 2 900 colleges and universities grant credit for CLEP exams **Gcse Physics Study Guide** Carol Tear,2007-09 This ultimate study guide with in depth GCSE course coverage is all you need for exam success Revise GCSE Physics has everything you need to achieve the GCSE grade you want It is written by GCSE examiners to boost learning and focus revision **The Hidden Curriculum - Faculty Made Tests in Science** Sheila Tobias,1997 *CLEP Official Study Guide* College Entrance Examination Board,2006-08 A guide to preparing for the thirty four different kinds of college CLEP exams that includes information on deciding which exams to take interpreting the scores taking the tests and other related topics *The Primary FRCA Structured Oral Examination Study Guide 1* Lara Wijayasiri,Kate McCombe,2025-03-25 The Primary FRCA Structured Oral Examination Study Guide 1 Physiology and Physics Third Edition provides answers to the questions regularly posed by the FRCA examiners The key information to succeed in the Primary FRCA examination has been collated and is delivered in a relevant and user friendly layout making exam preparation easier Fully updated and revised with new chapters covering Cleaning and decontamination

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Revise GCSE Science John Sadler,Ian Honeysett,Carol Tear,2006 Written by examiners and practising teachers this work offers study and homework support throughout GCSE It is useful as a reference source a lesson back up and as a revision guide

Official Study Guide for the CLEP Examinations, 1998 College Entrance Examination Board,1997 The College Level Examination Program CLEP tests are comprehensive 90 minute exams used to award full college credit for demonstrating college level achievement in a variety of areas in subjects This guide helps students decide which CLEP exams to take helps them interpret their CLEP scores and includes sample questions and answers for all 34 exams and a list of study resources

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CLEP Official Study Guide College Board,2001-08 This official guide developed by the sponsors of the CLEP exam includes a CD ROM for Windows that introduces test takers to the new computer based test and provides a tutorial for each exam

Guide to Science Teaching in Secondary Schools Archie L. Lacey, Virginia M. Rowley,1966

Revise GCSE Business Studies (2010 Exams Only) David Floyd,2005 New editions of the bestselling Revise GCSE Study Guides with a fresh new look and updated content in line with curriculum changes Revise GCSE contains everything students need to achieve the GCSE grade they want Each title has been written by a GCSE examiner to help boost students learning and focus their revision Each title provides complete curriculum coverage with clearly marked exam board labels so students can easily adapt the content to fit the course they are studying Revise GCSE is an ideal course companion throughout a student s GCSE study and acts as the ultimate Study

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