

QUESTION 1: MULTIPLE-CHOICE QUESTIONS

Various options are provided as possible answers to the following questions. Choose the answer and write only the letter (A-D) next to the question number (1.1 - 1.10) in the ANSWER BOOK, for example 1.5 D.

1.1 Which ONE of the following is a molecule with a multiple bond?

- A N_2
- B NH_3
- C OF_2
- D HOCl

(2)

1.2 The molecular formula of ethyne is C_2H_2 .

The shape of the ethyne molecule as predicted by the VSEPR theory is:

- A Trigonal planar.
- B Octahedral.
- C Linear.
- D Bent.

(2)

1.3 The table below indicates the boiling points of four liquids.

SUBSTANCE	BOILING POINT ($^{\circ}\text{C}$)
Water	100
Methylated spirits	78,5
Ethanol	78,37
Acetone	56

Which ONE of the above liquids has the lowest surface tension?

- A Water.
- B Methylated spirits.
- C Ethanol.
- D Acetone.

(2)

1.4 The predominant forces between the molecules in an ice crystal are called . . .

- A London forces.
- B Hydrogen bonding.
- C Induced dipole forces.
- D Polar covalent bonding.

(2)
[8]

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Daniel F McAuley



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1. Understanding the eBook Physical Science Past Exam Papers
 - The Rise of Digital Reading Physical Science Past Exam Papers
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Past Exam Papers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Past Exam Papers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Past Exam Papers
 - Personalized Recommendations
 - Physical Science Past Exam Papers User Reviews and Ratings
 - Physical Science Past Exam Papers and Bestseller Lists
5. Accessing Physical Science Past Exam Papers Free and Paid eBooks
 - Physical Science Past Exam Papers Public Domain eBooks

- Physical Science Past Exam Papers eBook Subscription Services
- Physical Science Past Exam Papers Budget-Friendly Options
- 6. Navigating Physical Science Past Exam Papers eBook Formats
 - ePub, PDF, MOBI, and More
 - Physical Science Past Exam Papers Compatibility with Devices
 - Physical Science Past Exam Papers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science Past Exam Papers
 - Highlighting and Note-Taking Physical Science Past Exam Papers
 - Interactive Elements Physical Science Past Exam Papers
- 8. Staying Engaged with Physical Science Past Exam Papers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Past Exam Papers
- 9. Balancing eBooks and Physical Books Physical Science Past Exam Papers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Past Exam Papers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Physical Science Past Exam Papers
 - Setting Reading Goals Physical Science Past Exam Papers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physical Science Past Exam Papers
 - Fact-Checking eBook Content of Physical Science Past Exam Papers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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