



NATIONAL
SENIOR CERTIFICATE

GRADE 11

NOVEMBER 2020

PHYSICAL SCIENCES P2
(CHEMISTRY)
EXEMPLAR

MARKS: 100
TIME: 1 hour



The question paper consists of 10 pages, including 4 extra sheets.
Use permanent marker for writing and drawing.

QUESTION 3

PHYSICAL SCIENCES PAPER 2

INTERMOLECULAR FORCES

GRADE 11

NOVEMBER 2020

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QUESTION 3 starts on a NEW page.

3.1 The boiling points in the table below were obtained during an investigation carried out to measure the boiling points of pure 1-hexanol. The table also lists the ΔH_{vap} and ΔH_{fusion} for 1-hexanol in kJ mol⁻¹ and 1 mol of 1-hexanol.

Compound	Molecular mass (M _r)	Boiling point (°C)
A	102	16,2
B	102	21

- 3.1.1 Define the term boiling point. (2)
- 3.1.2 Is pure 1-hexanol only composed of A and B? at 0 °C and 101 kPa of external pressure? (2)
- 3.1.3 Name the type of intermolecular force that exists between molecules of both compounds A and B and its relative value of these molecules. (2)
- 3.1.4 What kind of the compound A or B is 1-hexanol? Explain based on structural formula? (2)
- 3.1.5 What compound B or A is 1-hexanol? Explain based on structural formula? (2)
- 3.1.6 What compound B or A is 1-hexanol? Explain based on structural formula? (2)

Compound B (M_r 102) is shown in the table. Use a suitable molecular mass that both compounds A and B for the 1-hexanol boiling point of 16,2 °C.

- 3.1.6 Explain why the boiling point of compound B is 21 °C? (2)

3.2 Add table 3.2 is added to each column of pure 1-hexanol (100%) and water in separate test tubes as shown in the diagram below.



- 3.2.1 Is there any difference in the boiling points of the two liquids? (2)
- 3.2.2 Explain the process in QUESTION 3.2.1 about the boiling in the intermolecular forces involved. (2)

Examiner's use

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Physical Science Paper 2 Intermolecular Forces Investigation

Siddappa N.Byrareddy



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