



## PHYSICAL SCIENCE-11

### 3<sup>rd</sup> Quarterly Examination

Gulintolan National High School

S.Y. 2022-2023



Name: \_\_\_\_\_ Grade & Section: \_\_\_\_\_ Score: \_\_\_\_\_

#### I. MULTIPLE CHOICE

Direction: Encircle the letter of the correct answer. Strictly **ERASURE** means **WRONG**.

- Who concluded that galaxies are continuously moving away from one another?  
A. Ralph Alpher  
B. Edwin Hubble  
C. George Gamow  
D. Hans Bethe
- When was the universe created?  
A. Before the Big Bang  
B. After the Big Bang  
C. During the Big Bang  
D. During the expansion
- Which of the following elements is not formed during the stellar evolution?  
A. Neon  
B. Manganese  
C. Magnesium  
D. Silicon
- Which of the following best describe stellar nucleosynthesis?  
A. A process of producing gases in the stars.  
B. A process of producing heavy elements in the molecular cloud.  
C. A process of massive explosion in the stars.  
D. A process of forming heavy elements within the stars.
- In the main sequence star, what happen if the fusion stops?  
A. The energy in the core increases.  
B. The pressure in the core is reduce.  
C. The chemical reaction in the core stops.  
D. The molecular cloud in the core explodes.
- What is the absolute charge of a proton?  
A.  $+1.602176 \times 10^{-27}$   
B.  $-1.602176 \times 10^{-19}$   
C.  $+1.602176 \times 10^{-19}$   
D.  $-1.602176 \times 10^{-27}$
- What does Thomson discovered that every substance in this universe is made up of \_\_\_\_\_ from his experiments.  
A. neutrons  
B. protons  
C. electrons  
D. mass
- What is the lightest and smallest particles that's obtained from hydrogen(that is a positive ion).  
A. Electron  
B. Proton  
C. Neutron  
D Particle
- Which of the following statements about Rutherford's model of atom are correct?  
i) Considered the nucleus as positively charged.  
ii) Established that the  $\alpha$ -particles are four times as heavy as a hydrogen atom.  
iii) Can be compared to solar system. iv) Was in agreement with Thomson's model.  
A. i) and iii)  
B. ii) and iii)  
C. i) and iv)  
D. Only i)
- What is the type of bond in organic compounds?  
A. ionic bonds  
B. polar covalent bonds  
C. polar bonds  
D. covalent bonds
- What type of chemical bond holds the atoms of water molecule together?  
A. Hydrogen bond  
B. ionic bond  
C. Polar covalent bond  
D. Nonpolar covalent bond
- Which statement is true about the non-polar molecules?  
A. They are symmetric.  
B. They are asymmetric.  
C. They have lone pair on the central atom.  
D. They have charges
- Which one of the following molecules would have tetrahedral molecular geometry using the following compounds?  
A.  $\text{AsF}_5$   
B.  $\text{CF}_4$   
C.  $\text{XeF}_4$   
D.  $\text{NH}_3$
- Which best describe the shape and polarity of  $\text{CH}_4$ ?  
A. Tetrahedral polar  
B. Linear non-polar  
C. Tetrahedral non-polar  
D. Linear polar
- What is the molecular geometry of a  $\text{ClF}_3$  molecule?  
A. Triagonal pyramidal  
B. Triagonal Planar  
C. Tetrahedral  
D. T-shaped
- Which of the following atoms, when bonded to a fluorine atom, produces the highest degree of polarity?  
A. Sulphur  
B. Hydrogen  
C. Carbon  
D. Oxygen

# Physical Science Test Termgrade 1

**Anthony S. Fauci**



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