

4.2 Consider the equations for reactions I to III below.

A and B represent organic compounds that are POSITIONAL isomers.
X is an inorganic product.

I	$\text{CH}_3\text{CH}_2\text{CHCHCH}_3 + \text{HCl} \rightarrow \text{A} + \text{B}$
II	$\text{A} \xrightarrow[\Delta]{\text{H}_2\text{O}} \text{CH}_3\text{CH}_2\text{CH}_2\text{CH}(\text{OH})\text{CH}_3 + \text{X}$
III	$\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}(\text{OH})\text{CH}_3 \rightarrow \text{CH}_3\text{CH}_2\text{CHCHCH}_3 + \text{H}_2\text{O}$

Write down the:

4.2.1 Definition of positional isomers

(2)

4.2.2 Type of reaction represented by reaction I

(1)

4.2.3 STRUCTURAL formula of compound B

(3)

4.2.4 Formula of X

(1)

4.2.5

H₂O

(are they PO₁₀?)

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 - The Rise of Digital Reading Physical Science Chemistry Exemplar Paper 2 Memo
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Chemistry Exemplar Paper 2 Memo
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Chemistry Exemplar Paper 2 Memo
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Chemistry Exemplar Paper 2 Memo
 - Personalized Recommendations
 - Physical Science Chemistry Exemplar Paper 2 Memo User Reviews and Ratings
 - Physical Science Chemistry Exemplar Paper 2 Memo and Bestseller Lists
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 - Highlighting and Note-Taking Physical Science Chemistry Exemplar Paper 2 Memo
 - Interactive Elements Physical Science Chemistry Exemplar Paper 2 Memo
- 8. Staying Engaged with Physical Science Chemistry Exemplar Paper 2 Memo
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Chemistry Exemplar Paper 2 Memo
- 9. Balancing eBooks and Physical Books Physical Science Chemistry Exemplar Paper 2 Memo
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Chemistry Exemplar Paper 2 Memo
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Physical Science Chemistry Exemplar Paper 2 Memo
 - Setting Reading Goals Physical Science Chemistry Exemplar Paper 2 Memo
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physical Science Chemistry Exemplar Paper 2 Memo
 - Fact-Checking eBook Content of Physical Science Chemistry Exemplar Paper 2 Memo
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

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- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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