

Significant Figure and Scientific Notation Worksheet

1. Convert each of the following into scientific notation.

- | | |
|-------------------------|---------------------------|
| a) 3427 | k) 2205.2 |
| b) 0.00456 | l) 30.0×10^{-2} |
| c) 123,453 | m) 0.982×10^{-3} |
| d) 172 | n) 0.0473 |
| e) 0.000984 | o) 650,502 |
| f) 0.502 | p) 3.03×10^{-1} |
| g) 3100.0×10^2 | q) 20.4×10^5 |
| h) 0.0114×10^4 | r) 1000×10^{-3} |
| i) 107.2 | |
| j) 0.0000455 | |

2. Determine the number of significant digits in each of the following:

- | | | |
|-------------|--------------------------|---------------------------|
| a) 3427 | g) 3100.0×10^2 | m) 0.982×10^{-3} |
| b) 0.00456 | h) 0.0114×10^4 | n) 0.0473 |
| c) 123,453 | i) 107.2 | o) 650,502 |
| d) 172 | j) 0.0000455 | p) 3.03×10^{-1} |
| e) 0.000984 | k) 2205.2 | q) 20.4×10^5 |
| f) 0.502 | l) 30.0×10^{-2} | r) 1000×10^{-3} |

3. Convert each into decimal form.

- | | |
|---------------------------|----------------------|
| a) 1.56×10^4 | g) 6.9×10^4 |
| b) 0.56×10^{-2} | |
| c) 3.69×10^{-2} | |
| d) 736.9×10^5 | |
| e) 0.00259×10^5 | |
| f) 13.69×10^{-2} | |

Scientific Figures Mastering Chemistry Answers

Ian Pickup



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Table of Contents Scientific Figures Mastering Chemistry Answers

1. Understanding the eBook Scientific Figures Mastering Chemistry Answers
 - The Rise of Digital Reading Scientific Figures Mastering Chemistry Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Scientific Figures Mastering Chemistry Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Scientific Figures Mastering Chemistry Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Scientific Figures Mastering Chemistry Answers
 - Personalized Recommendations

- Scientific Figures Mastering Chemistry Answers User Reviews and Ratings
- Scientific Figures Mastering Chemistry Answers and Bestseller Lists
- 5. Accessing Scientific Figures Mastering Chemistry Answers Free and Paid eBooks
 - Scientific Figures Mastering Chemistry Answers Public Domain eBooks
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 - Highlighting and Note-Taking Scientific Figures Mastering Chemistry Answers
 - Interactive Elements Scientific Figures Mastering Chemistry Answers
- 8. Staying Engaged with Scientific Figures Mastering Chemistry Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Scientific Figures Mastering Chemistry Answers
- 9. Balancing eBooks and Physical Books Scientific Figures Mastering Chemistry Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Scientific Figures Mastering Chemistry Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Scientific Figures Mastering Chemistry Answers
 - Setting Reading Goals Scientific Figures Mastering Chemistry Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Scientific Figures Mastering Chemistry Answers
 - Fact-Checking eBook Content of Scientific Figures Mastering Chemistry Answers

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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