CBSE Sample Paper Class 6 Science Set 1

SUBJECT: SCIENCE CLASS: VI

MAX. MARKS: 80 DURATION: 2% HRS

General Instructions:

- All questions are compulsory. 610.
- This question paper contains 30 questions divided into four Sections A, B, C and D.
- (iii). Section A comprises 10 Fill in the Blanks Questions and 10 Multiple Choice Questions, each of 1 mark. Section B comprises of 8 questions of 2 marks each. Section C comprises of 8 questions of 3 marks each and Section D comprises of 4 questions of 5 marks each.

SECTION - A (1 mark each)

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- The habitat of organism that lives in water is called
- Birds are adapted for _____ mode of life.
- S.I. unit of mass is _____
- 4. A shadow cast by one heavenly object on another is called an
- A source of light which is of the size of the head of a common pin is called a of light.
- Combination of two or more cells is called a _____
- Hammering a magnet destroys its ______ character.
- 8. Wet clothes dry up in sun due to ______.
- 9. Increased humidity means increased amount of in the air.
- 10. Converting plant and animal wastes into manure is called _____

MULTIPLE CHOICE QUESTIONS: (10 marks)

- 11. Plant waste is given out in the form of thick fluid called
 - (a) gum

(b) urine

(c) latex

(d) sweat

- 12. Change in position of a body with time is called
 - (a) distance
- (b) motion
- (c) displacement
- (d) speed
- 13. We are able to see different objects around us through
 - (a) Their shadow (b) Regular reflection (c) Reflection (d) Irregular reflection
- 14. The tiny coiled metal wire present inside the bulb is called
 - (a) element (b) conductor
- (c) filament
- (d) none of these.

- In bulbs, electricity is converted into
 - (a) light energy

(b) heat energy

(c) sound energy

(d) mechanical energy

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15 TGT Social Science Test Papers EMRS Mocktime Publication, EMRS Exam Teachers TGT Social Science Test Papers 15 Practice Papers Tier 1 Eklavya Model Residential Schools as per Official Exam Pattern and Syllabus TETs Previous Years Papers (2021 - 2013) Mathematics and Science (Class 6-8) 2021 Arihant Experts, 2021-06-21 1 The book is complete practice capsule for CTET and TETs Entrances 2 Covers Previous Years Questions 2021 2013 of various Teaching Entrances 3 More than 3000 Questions are provided for practice 4 Well detailed answers help to understand the concepts Central Teacher Eligibility Test CTET or Teacher Eligibility Test TET are the national level teaching entrance exams that recruit eligible candidates as teacher who are willing to make their careers in the stream of teaching at Central or State Government Schools Prepared under National curriculum pattern the current edition of CTET TETs Previous Years Solved Papers Mathematics Science for Class 6 8 is a complete practice package for teaching entrances This book covers all the previous years questions 2021 2013 providing complete detailed explanations of each question It has more than 3000 Questions that are asked in various Teaching Entrances which promote self evaluation by enabling not just practicing and revising concepts but also to keep track of self progress Well detailed answers help students to win over doubt and fears associated with exam Preparation done from this book proves to be highly useful for CTET Paper II in achieving good rank TABLE OF CONTENT Solved Paper 2021 2013 Oswaal NTA CUET (UG) 5 Mock Test Papers Computer Science/Informatics Practices (For 2025 Exam) Oswaal Editorial Board, 2024-09-05 The National Testing Agency NTA under the directive of the Ministry of Education and the UGC has been entrusted with conducting the Common University Entrance Test CUET for admissions into undergraduate programs at Central Universities under the Ministry of Education This test is the gateway for admission into undergraduate programmes at Central Universities under the Ministry of Education as well as other participating universities institutions organizations and autonomous colleges The CUET UG curriculum is based on the syllabus issued by NTA CUET UG scores are mandatory required while admitting students to undergraduate courses in 283 Central States and other participating universities institution organisations for the Academic Session 2024 25 The MCQ based hybrid question paper will include language specific domain and general topics sections Participating universities organizations will prepare a merit list and may conduct individual counselling based on the CUET UG scorecard provided by the NTA Oswaal CUET UG Sample Question Paper is your strategic companion designed to elevate your performance and simplify your CUET journey for success in this computer based test Here's how this book benefits you Valuable Exam Insights with Latest Solved Paper 2024 Extensive Practice with 500 approx Questions Concept Clarity with 250 Explanations Expert Tips to crack the exam in 1st Attempt In 2024 nearly 15 lakh candidates registered for CUET UG Though the test may feel challenging the right preparation and resources can help you secure a top rank With dedication and the right tools you can excel and gain admission to your preferred Central University Best of luck let these Mock Papers be

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contest general awareness and mental ability test in every monthly issue **Resources in Education** ,1998-10 The Effective Teaching of Secondary Science John Parkinson, 2014-06-03 The Effective Teaching of Secondary Science encourages the trainee teacher to develop effective skills for teaching science to secondary school pupils The comprehensive coverage of topics and issues provides good foundations for trainee teachers who are encouraged to test and evaluate different techniques Practical advice is offered in areas such as lesson planning the preparation of worksheets planning practical activities and safety in the laboratory The book also discusses the use of information technology as well as multicultural and gender issues and the teaching of pupils with special needs Much of the work covered is undepinned by areas of educational research such as educational theory and psychology and sociology of education Information on the requirements of the national curriculum and on post 16 science courses is given and includes a number of assessment techniques for the problematic area of assessing science attainment target 1 Spotlight Science Keith Johnson, Sue Adamson, Gareth Williams, 2000 Topic Outlines show parts of the PoS to be covered the relationship of the topic to aspects of KS2 and KS4 and warn of equipment that may need special preparation time in advance Topic Maps are provided for students Lesson Notes relating to each double page spread in the students book offer objectives ideas for each lesson detailed references to the PoS level descriptions safety points with references to CLEAPPS HAZCARDS ICT support cross curricular links and equipment lists Answers to all questions in the students book are also provided Additional support material provide Homework Sheets Help and Extension Sheets to optimise differentiation Sc1 Sc1 Skill Sheets Thinking about activities to improve integration of CASE activities with Spotlight Science Revision Quizzes and Checklists etc Extra Help Sheets for each topic extend the range of support for Sc1 and Sc2 4 Challenge Sheets for each topic provide a variety of enrichment activities for more able students. They consist of a variety of challenging activities which will present students with opportunities to develop problem solving thinking presentational and interpersonal skills Technician s Cards include help to prepare lessons equipment requirements and CLEAPPS HAZCARD references For more information visit the website at www spotlightscience co uk Notabilia and test papers on Plato, Gorgias ,1894

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web m x c x Δt equation 1 where m is the total mass of the solution solute plus solvent c is the specific heat of the solution and Δt is the observed temperature change the specific heat of the solution is generally assumed to be the same as that of quided inquiry designing the ideal hand warmer - Dec 27 2022

web the heat of reaction Δ hsoln is written after the products in units of kj mol rxn for example sodium hydroxide dissolves exothermically releasing 44 2 kilojoules per mole dissolved this process is represented as naoh s na aq oh aq Δ hsoln 44 2 kj mol rxn write equations to similarly represent the dissolving process for

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web hand warmers are usually placed inside the gloves to keep your hands warm the heat keeping your hands warm is formed as a result of an exothermic reaction which occurs in a small packet hand warmers have 5 main ingredients in them salt carbon polypropylene vermiculite water and iron

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web background the hand warmers are filled with sodium acetate click the metal disc and you will notice the liquid changes state and becomes a solid in the process heat is released the sodium acetate is a supersaturated solution

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web oct 9 2021 sodium acetate coin or other small piece of metal plastic bags seal the coin inside a small bag place this bag inside a larger bag containing supersaturated sodium acetate get this by dissolving as much sodium acetate as possible in hot water this chemical hand warmer is reusable

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web the goal of my experiment and research is to evaluate the various number of compounds to be used as hand warmers and locate the ones that can raise the temperature to 20 degrees celsius in 100 g water and costs as little as possible

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web in this lab you are challenged to use chemistry to determine which ionic compound makes the best hand warmer in this experiment you will collect data that will allow you to calculate the heat of solution Δh soln or the amount of energy released or absorbed in kj mol of the substance being dissolved

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web mar 4 2014 contains water the hand warmer is struck in a manner that ruptures the inner pouch releasing the ionic salt into the water of the outer pouch the salt dissolves and the water warms this type of hand warmer tends to produce a more vigorous heat than the dry powder type of hand warmer but does not produce heat for quite as long

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web hand warmers provide a unique and fun way to study the chemistry of crystal formation and heat generation by pressing a button in a pouch which contains a supercooled solution you start a rapid exothermic heat producing crystallization in this science fair project you will determine how the starting temperature affects hand warmer chemistry

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web lab investigation in today s experiment you will design and execute an experimental procedure to determine which of three ionic compounds is most suitable for use in a hand warmer the three compounds you will use is determined by the group number you are assigned the ideal hand warmer

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