

- The entropy change of the surroundings is driven by heat flow.
 - 2nd law: every energy transfer or transformation increases the entropy of the universe.
- c. Conservation of matter in chemical systems.
- Matter neither created nor destroyed, so we must have the same number and type of atoms after the chemical change as were present before the chemical change.
- d. Kinetic and potential energy
- **Kinetic energy:** energy in motion.
 - **Potential energy:** the energy possessed by a body by virtue of its position relative to others, stresses within itself, electric charge, and other factors.
- e. Transformations between different forms of energy (thermal, chemical, radiant, nuclear, mechanical, electrical, electromagnetic, sound, and chemical).
- **Energy Transformations Examples:** An energy transformation is the change of energy from one form to another. Energy transformations occur everywhere every second of the day. There are many different forms of energy such as electrical, thermal, nuclear, mechanical, electromagnetic, sound, and chemical.

f. Differences between chemical and physical properties/changes

- The difference between a physical reaction and a chemical reaction is composition.
- **Chemical property/changes:** The changing of make of a substance is not necessarily an indication of a chemical change. For example, changing the color of a metal does not change its physical properties; however, in a chemical reaction, a color change is usually an indication that a reaction is occurring.
 - Change in Temperature
 - Change in Color
 - Noticeable Odor (after reaction has begun)
 - Formation of a Precipitate
 - Formation of Bubbles
- **Physical property/changes:** A physical change there is a difference in the appearance, smell, or simple display of a sample of matter without a change in composition.
 - Texture
 - Color
 - Temperature
 - Shape
 - Change of State (Boiling Point and Melting Point are significant factors in determining this change.)

Physical properties include many other aspects of a substance. The following are (but not limited to) physical properties:

- Luster
- Malleability
- Ability to be drawn into a thin wire
- Density
- Viscosity
- Solubility
- Mass
- Volume

g. Various temperature scales (Celsius, Fahrenheit, Kelvin)

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LL Leslie

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1. Understanding the eBook Praxis Study Guide Science
 - The Rise of Digital Reading Praxis Study Guide Science
 - Advantages of eBooks Over Traditional Books
2. Identifying Praxis Study Guide Science
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Praxis Study Guide Science
 - User-Friendly Interface
4. Exploring eBook Recommendations from Praxis Study Guide Science
 - Personalized Recommendations
 - Praxis Study Guide Science User Reviews and Ratings
 - Praxis Study Guide Science and Bestseller Lists
5. Accessing Praxis Study Guide Science Free and Paid eBooks
 - Praxis Study Guide Science Public Domain eBooks
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 - Praxis Study Guide Science Budget-Friendly Options
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- Praxis Study Guide Science Enhanced eBook Features
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 - Highlighting and Note-Taking Praxis Study Guide Science
 - Interactive Elements Praxis Study Guide Science
- 8. Staying Engaged with Praxis Study Guide Science
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Praxis Study Guide Science
- 9. Balancing eBooks and Physical Books Praxis Study Guide Science
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Praxis Study Guide Science
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Praxis Study Guide Science
 - Setting Reading Goals Praxis Study Guide Science
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Praxis Study Guide Science
 - Fact-Checking eBook Content of Praxis Study Guide Science
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
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