# How to Balance a Chemical Equation

Reactants

 $N_2 + H_2$ 

**Atom Counts** 

Nitrogen: 2

From the Subscript

Hydrogen: 2

Product
NH-

**Atom Counts** 

Nitrogen: 1

Hydrogen: 3



When balancing a chemical equation, there should be equal numbers of every atom on both sides of the equation

 $N_2 + 3H_2 \longrightarrow 2NH_3$ 

Nitrogen: 2

Hydrogen: 3\*2=6

From the Coefficient

Nitrogen: 2\*1=2

Hydrogen: 2\*3=6



## **Practice Reaction And Equation**

Neil D. Jespersen, Pamela Kerrigan

#### **Practice Reaction And Equation:**

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