

Name \_\_\_\_\_ Class \_\_\_\_\_ Date \_\_\_\_\_

## Assessment

## Chapter Test B

## Atomic Physics

## MULTIPLE CHOICE

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- \_\_\_\_\_ 1. What is the frequency of a photon with an energy of  $1.00 \times 10^{-19} \text{ J}$ ?  
( $h = 6.63 \times 10^{-34} \text{ J}\cdot\text{s}$ )  
a.  $1.00 \times 10^{14} \text{ Hz}$  c.  $3.00 \times 10^{12} \text{ Hz}$   
b.  $2.00 \times 10^{14} \text{ Hz}$  d.  $4.00 \times 10^{12} \text{ Hz}$
- \_\_\_\_\_ 2. Light with an energy equal to three times the work function of a given metal causes the metal to eject photoelectrons. What is the ratio of the maximum photoelectron kinetic energy to the work function?  
a. 1 : 1 c. 3 : 1  
b. 2 : 1 d. 4 : 1
- \_\_\_\_\_ 3. A monochromatic light beam with a quantum energy value of 3.0 eV is incident upon a photocell. The work function of the photocell is 1.6 eV. What is the maximum kinetic energy of the ejected electrons?  
a. 4.6 eV c. 1.4 eV  
b. 4.8 eV d. 2.4 eV
- \_\_\_\_\_ 4. What causes the bright lines in the emission spectrum of an element to occur?  
a. Photons are absorbed when electrons jump from a higher-energy to a lower-energy state.  
b. Photons are emitted when electrons jump from a higher-energy to a lower-energy state.  
c. Photons are absorbed when electrons jump from a lower-energy to a higher-energy state.  
d. Photons are emitted when electrons jump from a lower-energy to a higher-energy state.
- \_\_\_\_\_ 5. What causes the dark lines in the absorption spectrum of an element to occur?  
a. Photons are absorbed when electrons jump from a higher-energy to a lower-energy state.  
b. Photons are emitted when electrons jump from a higher-energy to a lower-energy state.  
c. Photons are absorbed when electrons jump from a lower-energy to a higher-energy state.  
d. Photons are emitted when electrons jump from a lower-energy to a higher-energy state.

# Physics Chapter Test

**Joy Paul Guilford, John I. Lacey**



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### **Table of Contents Physics Chapter Test**

1. Understanding the eBook Physics Chapter Test
  - The Rise of Digital Reading Physics Chapter Test
  - Advantages of eBooks Over Traditional Books
2. Identifying Physics Chapter Test
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Physics Chapter Test
  - User-Friendly Interface
4. Exploring eBook Recommendations from Physics Chapter Test
  - Personalized Recommendations
  - Physics Chapter Test User Reviews and Ratings
  - Physics Chapter Test and Bestseller Lists
5. Accessing Physics Chapter Test Free and Paid eBooks

- Physics Chapter Test Public Domain eBooks
- Physics Chapter Test eBook Subscription Services
- Physics Chapter Test Budget-Friendly Options
- 6. Navigating Physics Chapter Test eBook Formats
  - ePub, PDF, MOBI, and More
  - Physics Chapter Test Compatibility with Devices
  - Physics Chapter Test Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Physics Chapter Test
  - Highlighting and Note-Taking Physics Chapter Test
  - Interactive Elements Physics Chapter Test
- 8. Staying Engaged with Physics Chapter Test
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Physics Chapter Test
- 9. Balancing eBooks and Physical Books Physics Chapter Test
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Physics Chapter Test
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Physics Chapter Test
  - Setting Reading Goals Physics Chapter Test
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physics Chapter Test
  - Fact-Checking eBook Content of Physics Chapter Test
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development



- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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