

## Physics 1320: Study Guide Ch 21

### Charges and Electric Forces

#### OBJECTIVES:

- a) To understand the nature of electric charges.
- b) To understand Coulomb's Law.

#### GLOSSARY:

**Electric charge:** May be positive or negative. Generally denoted by  $q$  for small charges and  $Q$  for large charges on macroscopic objects. Units of coulombs [C]. The fundamental charge element is  $e = 1.602 \times 10^{-19}$  C. This is the magnitude of the charge of an electron ( $-e$ ) and a proton ( $+e$ ).

**Point charge:** A charge that exists at a point in space but has no size. It does have an electric charge and possibly a mass.

**Continuous charge:** An object (with some size) for which electric charge is spread out over the object.

**Valence electrons:** The electrons in the outer shell of an atom.

**Conductor:** An object whose valence electrons are free to move due to an electric force. Positive charges (protons) are in the nucleus of atoms and do not move.

**Insulator:** An object whose valence electrons are NOT free to move. All electrons in an insulator are bound to their atom.

**Induction:** Movement of charge due to bringing a charged object *near* another object.

**Conduction:** Movement of charge due to bringing a charged object *to touch* another object. When the charge redistributes between the two objects, the total charge stays the same.

#### COULOMB FORCE LAW:

The magnitude of the force between two charges:

$$F = k_e \frac{|q_1||q_2|}{r^2}$$
 where  $r$  is the distance between  $q_1$  and  $q_2$ . The force is attractive if the charges are opposite, repelling if the charges are the same.

$$k = \frac{1}{4\pi\epsilon_0} = 8.99 \times 10^9 \frac{\text{Nm}^2}{\text{C}^2}$$

# Physics 21 Study Guide

**Thomas Griffiths**



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