

Quantum Theory

- a. Energy is quantized in $\frac{h\nu}{2\pi}$ and $\frac{h\nu}{2\pi}$ is the minimum energy
- b. Photo electron of energy is emitted if $\frac{h\nu}{2\pi} > \phi$
- c. In quantized light energy is emitted $\frac{h\nu}{2\pi}$
- d. The energy of a photon of light depends on the $\frac{h\nu}{2\pi}$ of light

$$E = \frac{h\nu}{2\pi} \text{ constant}$$

_____ now the question is how to indicate this phenomenon

- a. Light energy of $\frac{h\nu}{2\pi}$ is light having a certain amount of energy depending on the $\frac{h\nu}{2\pi}$ $h\nu = E_1 + E_2$
- b. Since $\frac{h\nu}{2\pi}$ is constant a single photon of light
- c. As also the constant amount of $\frac{h\nu}{2\pi}$ is given out at this point
- d. At point $\frac{h\nu}{2\pi}$ and forward energy jumps, this electron will be ejected
- e. At a quantized energy jumps that electron needs the energy $h\nu$ and the electron will be $\frac{h\nu}{2\pi}$ and back to zero level

What happens for light of constant light

(a)

(b)

Quantum Theory: Photoelectric Effect $\frac{h\nu}{2\pi}$

Light energy of $\frac{h\nu}{2\pi}$ is light having a certain amount of energy depending on the $\frac{h\nu}{2\pi}$ and the electron will be $\frac{h\nu}{2\pi}$ and back to zero level

Physics 1302 Note Taking Guide Answers

Lauren Gardner



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