#### PHYSICAL SCIENCES GRADE 11

#### TERM 1: PHYSICS EXPERIMENT

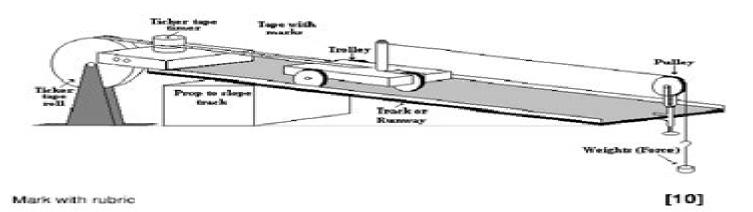
#### KNOWLEDGE AREA: MECHANICS

#### VERIFICATION NEWTON'S SECOND LAW OF MOTION

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NB: All diagrams, data and graphs are just examples. Mark according to the experiments as conducted by the learners at your school.

#### PART 1



# **Physical Sciences Pscope Grade11**

# **Mildred Patten**

# **Physical Sciences Pscope Grade11:**

The Art of Teaching Science Jack Hassard, Michael Dias, 2013-07-04 The Art of Teaching Science emphasizes a humanistic experiential and constructivist approach to teaching and learning and integrates a wide variety of pedagogical learning tools These tools involve inquiry and experimentation reflection through writing and discussion as well as experiences with students science curriculum and pedagogy Becoming a science teacher is a creative process and this innovative textbook encourages students to construct ideas about science teaching through their interactions with peers professionals and instructors and through hands on minds on activities designed to foster a collaborative thoughtful learning **Physical Science Experiments** Pam Walker, Elaine Wood, 2010 Presents new tested experiments related to environment. the intriguing field of physical science The experiments are designed to promote interest in science in and out of the classroom and to improve critical thinking skills Students Learning Science Christine Y. O'Sullivan, Andrew R. Weiss, Janice M. Askew, 1998 This report on teachers academic preparation and professional development the amount of emphasis science instruction receives in schools student course taking and the availability of school resources that support science learning is intended primarily for policy makers school administrators and educators concerned with state or school level policies Data is drawn from the 1996 National Assessment of Educational Progress NAEP and results are presented using the students as the unit of analysis Appendises present an overview of procedures used for the NAEP 1996 Science Assessment and standard errors Contains 14 figures and 25 tables DDR Students learning science : a report on policies and practices in U.S. schools, Qualification Standards for Positions Under the General Schedule United States. Office of Personnel Management. Occupational Standards Branch, 1979 Resources in Education ,2001-04 Forensic Science Experiments Pam Walker, Elaine Wood, 2010-05-12 Provides twenty experiments in Education ,1973 forensic science that will intrigue both students and teachers and promote the interest in multiple science process skills

**Forum** ,1993 A Guide to Curriculum Development in Science ,1986 Designed to facilitate development of the local school district curriculum Provides quidelines for conducting a curriculum assessment ENC Focus .2001 School Discover Science: Teacher's annotated edition, 1991 Science content helps develop the Science and Mathematics ,1914 skills needed to understand how science works learn new concepts solve problems and make decisions in today s The Sourcebook for Teaching Science, Grades 6-12 Norman Herr, 2008-08-11 The Sourcebook for technological society Teaching Science is a unique comprehensive resource designed to give middle and high school science teachers a wealth of information that will enhance any science curriculum Filled with innovative tools dynamic activities and practical lesson plans that are grounded in theory research and national standards the book offers both new and experienced science teachers powerful strategies and original ideas that will enhance the teaching of physics chemistry biology and the earth and space sciences Science Blogging Christie Wilcox, Bethany Brookshire, Jason G. Goldman, 2016-03-01 Here is the essential

how to guide for communicating scientific research and discoveries online ideal for journalists researchers and public information officers looking to reach a wide lay audience Drawing on the cumulative experience of twenty seven of the greatest minds in scientific communication this invaluable handbook targets the specific questions and concerns of the scientific community offering help in a wide range of digital areas including blogging creating podcasts tweeting and more With step by step guidance and one stop expertise this is the book every scientist science writer and practitioner needs to approach the Wild West of the Web with knowledge and confidence School Life ,1945 The Science Teacher ,2009

Salaries of Federal Government Employees. Hearings ... on S. 558, S. 559, S. 1762, S. 1772, S. 1790 ... May 9, 11, 18, 20, and 25, 1949 United States. Congress. Senate. Committee on Post Office and Civil Service, 1949 New Horizons in Mathematics and Science Education ,2001 **Educational and Psychological Research** Mildred Patten, 2016-09-13 This edition of our popular reader is entirely updated with timely articles that deal with broad issues in education and psychology For a research reader with emphasis on classroom issues see A Cross Section of Educational Research This collection shows students the many inherent weaknesses of empirical methods as well as models of excellence Unlike research readers that contain made up articles this one has both face validity and content validity Your students will appreciate reading real articles with real results The 34 research articles illustrate qualitative research combined qualitative quantitative research content documentary analysis correlational research test validity and reliability research causal comparative research true experimental research guasi experimental research pre experimental research program evaluation meta analysis The lines are numbered sequentially making it easy to refer to specific parts of each article during classroom discussions This single source of research articles is convenient for you and your students Easy to coordinate with any research methods text this reader is an ideal way to make research methods come alive Field tested for student interest and comprehension All major methods of research are illustrated with real research articles drawn from a wide variety of journals See Table of Contents Your students will become familiar with a wide variety of writing and organizational styles that real researchers use Despite their variety all articles in this book are highly comprehensible Factual Questions at the end of each article encourage students to read for mythological points they might otherwise overlook Questions for Discussion encourage students to address broad issues of research design An instructor's answer key will be shipped with your examination copy The interesting research topics will make your students look forward to doing their research reading homework

The book delves into Physical Sciences Pscope Grade11. Physical Sciences Pscope Grade11 is a vital topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Physical Sciences Pscope Grade11, encompassing both the fundamentals and more intricate discussions.

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web now realizing that heating and cooling curves represent the amount of heat absorbed or released by a substance during phase changes remember heat uses the variable q here we have a heating curve versus a cooling curve in a heating curve our substance is absorbing heat

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web nov 15 2021 for the heating and cooling curve of any given substance the solid phase of that substance will be represented in the lower left corner of the graph where the temperature is at its lowest

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web describe the processes represented by typical heating and cooling curves and compute heat flows and enthalpy changes accompanying these processes explain the construction and use of a typical phase diagram

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web heating and cooling curves a heating curve can be used to calculate the enthalpy when a substance is heated if we were to heat 25 00 g of water from 15 0 c to 115 0 c we can determine  $\Delta h$  for the heating process

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web 11 12 heating and cooling curves when a substance isolated from its environment is subjected to heat changes corresponding changes in temperature and phase of the substance is observed this is graphically represented by heating and cooling curves

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web determine the final temperature and state of the ammonia once the system reaches equilibrium assuming that no heat is lost to the surroundings use the information below normal boiling point of nh 3 33 3 c  $\Delta h$  vap 23 4 kj mol specific heat nh 3 l

4 74 j g k specific heat nh 3 g 2 06 j g k

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web apr 3 2023 the experiment described above can be summarized in a graph called a heating curve figure below figure 11 16 1 11 16 1 in the heating curve of water the temperature is shown as heat is continually added changes of state occur during plateaus because the temperature is constant

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web mar 29 2013 heating curves imagine that you have a block of ice that is at a temperature of 30 c well below its melting point the ice is in a closed container as heat is steadily added to the ice block the water molecules will begin to vibrate faster and faster as they absorb kinetic energy

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web the cooling curve a plot of temperature versus cooling time in figure pageindex 4 plots temperature versus time as a 75 g sample of steam initially at 1 atm and 200 c is cooled although we might expect the cooling curve to be the mirror image of the heating curve in figure pageindex 3 the cooling curve is not an

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web once all of the material has solidified at the time indicated by point c the cooling will continue at a rate determined by the heat capacities of the two solids a and b the composition and of course the geometry of the experimental set up by measuring cooling curves for samples of varying composition one can map the entire phase

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web for water this temperature is 100 c because the boiling point for water is 100 c different substances have different melting points and boiling points but the shapes of their heating curves are very similar for example this is the heating curve for iron a metal that melts at 1538 c and boils at 2861 c cooling curves

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web list of government bursaries in south africa there are many departments within the
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web state one reason in terms of energy to support your answer answer endothermic the products have more energy than the
reactants b on the diagram provided in your answer booklet draw a dashed line to indicate a potential energy curve for the
reaction if a catalyst is added 46 given the potential energy diagram for a reaction
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web when the reactants have more potential energy than the products have and the extra energy is released to the

surroundings it is called activation energy the height of the hill on a potential energy diagram is a measure of the endothermic the reaction above can be categorized as

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web this is like a one dimensional system whose mechanical energy e is a constant and whose potential energy with respect to zero energy at zero displacement from the spring s unstretched length x 0 is u x 12 1 2 kx 2 figure 8 5 2 8 5 2 a a glider between springs on an air track is an example of a horizontal mass spring system

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web 1 if the pe of the reactants is 430kj and the pe of the products is 190kj what is the  $\Delta h$  2 if the pe of the reactants is 430kj and the activation energy is 220 how much total energy will be needed to be added for the reaction to proceed transcribed image text potential energy diagram answer the questions using the graph

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web figure 5 5 1 a potential energy diagram shows the total potential energy of a reacting system as the reaction proceeds a in an endothermic reaction the energy of the products is greater than the energy of the reactants and  $\Delta$  h is positive b in an exothermic reaction the energy of the products is lower than the energy of the

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web pertanyaan serupa iklan perhatikan diagram energi potensial berikut pada diagram di atas energi aktivasi untuk reaksi baliknya ditandai dengan

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web mar 17 2016 3 what is the activation energy of a reaction and how is this energy related to the activated complex of the reaction ea is the minimum amount of energy for a reaction to occur it is the amount of energy required to create an activated complex 4 what happens when a catalyst is used in a reaction a catalyst changes the reaction

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web aug 8 2022 the energy changes that occur during a chemical reaction can be shown in a diagram called a potential energy diagram or sometimes called a reaction progress curve a potential energy diagram shows the change in potential energy of a system as reactants are converted into products the figure below shows basic potential energy potential energy article energy khan academy - Apr 17 2022

web key points potential energy is energy that has the potential to become another form of energy an object s potential energy depends on its physical properties and position in a system potential energy comes in many forms such as gravitational potential energy due to an object s mass and position in a gravitational field answered use the potential energy diagram shown bartleby - Jul 21 2022

web the energy of the reactants is shown on the energy diagram above on the right side of the energy diagram draw a horizontal line segment to indicate the energy of the products draw a vertical line segment to indicate  $\Delta h$  for the reaction and label it with the correct value see attached image

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web aug 16 2019 1 answer sorted by 4 the activation energy is not the average kinetic energy that the colliding particles need in order to successfully react instead it is the energy barrier between reactants and products often the energy required to cross the barrier comes from the kinetic energy of collisions

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web all questions are answered herein chemistry tutorial 9 02b potential energy diagrams how to draw and label pe diagrams for exothermic and endothermic reactions and what effect adding a catalyst or inhibitor has on the diagram

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web may 1 2013 a potential energy diagram shows the change in potential energy of a system as reactants are converted into products figure below shows basic potential energy diagrams for an endothermic a and an exothermic b reaction recall that the enthalpy change  $\Delta$  h is positive for an endothermic reaction and negative for an

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time ms 0 00 1 00 2 00 3 00 4 00 5 00 6 00 7 00 8 00 9 00 10 0 potential energy kcal 0 00 10 0 **energy diagrams chemistry libretexts** - Sep 22 2022

web purdue chem 26100 organic chemistry i wenthold chapter 5 the study of chemical reactions chemistry 12 worksheet 1 2 potential energy diagrams tsfx - Aug 02 2023 web use the potential energy diagram to answer the questions below 1 is the overall reaction as shown exothermic or endothermic 2 what is the activation energy for the forward reaction 3 what is the